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African Journal of Food Science

Table of Content: Volume 12 Number 8 August 2018

ARTICLES

- Antioxidants activity of the cyanobacterium, *Arthrospira (Spirulina) fusiformis* cultivated in a low-cost medium** 188
Angelina Michael, Margareth S. Kyewalyanga, Matern S. Mtolera and Charles V. Lugomela
- Nutritional potential of underutilized gum arabic tree seeds (*acacia nilotica*) and locust bean seeds (*Parkia biglobosa*)** 196
Ajayi K., Adepoju O. T., Taiwo O. M., Omojola S. T. and Aladetuyi M. E.
- Physicochemical, microbiological and sensory characteristics of cashew milk formulated yoghurt** 204
Jayeola Olayinka, Yahaya Eugene, Ogunwolu Olalekan, Igbinadolor Richard and Mokwunye Chuka

Full Length Research Paper

Antioxidants activity of the cyanobacterium, *Arthrospira (Spirulina) fusiformis* cultivated in a low-cost medium

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***Spirulina* species are known to have a good nutritional profile and antioxidant properties against reactive oxygen species. However, little is known about the antioxidant contents and the scavenging ability of *Arthrospira fusiformis*, cultivated under various conditions. This study aimed at evaluating the content of antioxidants (total phenols, total flavonoids, β -carotene, and lycopene) and the activity of *A. fusiformis* produced using low-cost culture (LCMA) and standard culture (Zarrouk) media. The results revealed that *A. fusiformis* is rich in antioxidants and it possesses high scavenging and chelating activities. Interestingly, the LCMA was superior over the Zarrouk medium as it resulted in spirulina with a higher amount of antioxidants and lower EC₅₀ values. In this context, production of natural antioxidants can be maximized through the use of cost-saving, inorganic culture medium.**

Key words: *Arthrospira fusiformis*, spirulina, total phenols, total flavonoids, carotenoids, scavenging activity, low-cost culture (LCMA) medium, Zarrouk medium.

INTRODUCTION

Arthrospira, commonly known as spirulina is a filamentous helical shaped cyanobacterium belonging to the family Oscillatoriaceae (Kumar et al., 2005; Rasool et al., 2006). It occurs naturally in warm alkaline lakes of the tropical and sub-tropical countries (Germán Chamorro-Cevallos and Vázquez-Sánchez, 2008; Habib et al., 2008; Shalaby and Shanab, 2013; Kumari et al., 2015). Spirulina is proven to be toxicologically free (Germán Chamorro-

Cevallos and Vázquez-Sánchez, 2008; Gutiérrez-Salmeán et al., 2015) and it has been cultivated massively in several countries especially those in the Asian and American continent and used as protein and vitamins supplement in the diets (Rasool et al., 2006; Belay, 2008; Salamatullah, 2014). Spirulina is a rich source of protein (about 50 to 70%), essential amino acids, vitamins, minerals and unsaturated fatty acid (Pandey et al., 2010;

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Gutiérrez-Salmeán et al., 2015). More interestingly, it possesses the antioxidant and antiradical properties being attributed by phytonutrients such as phenolics, phycocyanin, tocopherol and β -carotene (Colla et al., 2007; Shalaby and Shanab, 2013; Al-Dhabi and Valan Arasu, 2016; Ismaiel et al., 2016). Thus, consumption of spirulina improved the resistance of consumers against oxidative stress. Several studies have pointed out the efficiency of spirulina as an anti-viral and anticancer (Kumar et al., 2005), anti-inflammation (Rasool et al., 2006), and anti-allergic and antibacterial (Belay, 2008). It is further reported that the antioxidant activity in spirulina extract against lipid peroxidation is even more powerful than that of synthetic antioxidants such as butylated hydroxyanisole (BHA) and butylated hydroxytoluene (BHT) (Chopra and Bishnoi, 2008; Tarko et al., 2012).

Among the *Arthrospira* species, *Arthrospira platensis* and *Arthrospira maxima* are the most studied and cultivated for human food, dietary supplement, and animal feed additive (Belay, 2008). There is scarce information regarding the potential of *Arthrospira fusiformis* especially the nutritional and bioactive composition. However, some studies have confirmed the anti-inflammatory and anti-cancer properties of *A. fusiformis* (Mathew et al., 1995; Rasool et al., 2006; Deng and Te-Jin, 2010). More recently, Mulokozi (2016) suggested that the cultivated *A. fusiformis* can replace a significant amount of the fishmeal in tilapia feeds. Nevertheless, antioxidant and nutritional contents may vary due to factors such as culture conditions, culture media, analysis methods, type and source of the organism (Habib et al., 2008; Gutiérrez-Salmeán et al., 2015; Al-Dhabi and Valan Arasu, 2016).

Spirulina can be cultivated under laboratory conditions as well as outdoor for large-scale systems. The outdoor culture systems rely mainly on Zarrouk medium (Belay, 2008; Madkour et al., 2012; Tarko et al., 2012) though it is highly expensive. Thus, efforts have been made to develop a more convenient and a less expensive culture media (Raouf et al., 2006; Chen, 2011; Gami et al., 2011; Madkour et al., 2012), which can produce high-quality spirulina biomass comparable to the standard culture medium. Cultivation trials of spirulina conducted in Tanzania used the culture medium termed as OFER, which composes fewer analytical grade chemicals as compared to those of Zarrouk medium (Mulokozi, 2016). In joining the effort to reduce the cost of production and maximizing spirulina biomass, in this study a new culture medium, namely, LCMA was introduced. The LCMA was formed by mixing a low-cost inorganic fertilizer (NPK10-20-20), two analytical grade chemicals from Zarrouk medium (sodium chloride and sodium bicarbonate) and drops of trace element solution. Moreover, the aim of this study was to assess the antioxidant components (total phenols, total flavonoids, β -carotene, and lycopene) and antioxidant properties in the fresh and dried extracts of *A. fusiformis* cultivated in the cost-effective medium, and

compared the results to the standard culture medium.

MATERIALS AND METHODS

Preparation of culture media and spirulina cultivation

The strain of *A. fusiformis* used in this study was obtained from the stock kept at the Institute of Marine Sciences, University of Dar es Salaam, Tanzania. The algal sample was previously collected from Lake Big Momela, Tanzania, and purified according to Mulokozi (2016). The stock was maintained in 2000 ml conical flasks in standard culture medium (Zarrouk). On the beginning of this study, spirulina was cultivated in two synthetic media (Table 1) whereby Zarrouk was used as a standard medium (Kumari et al., 2015) and LCMA as an alternative low-cost medium for mass culture. NPK10-20-20 fertilizer was obtained from authorized dealers of agricultural inputs farmers in Tanzania at Kariakoo Market. The analytical grade chemicals for Zarrouk medium were purchased from laboratory equipment and chemical supplier in Zanzibar (Zan-Lab Equipment).

The experiment was carried out in 10 L aquaria containing 1900 ml of culture media and 100 ml (0.038 g/L dry weight) of spirulina. Three aquaria were set for each Zarrouk and LCMA. The culture was incubated for 30 days in a growth chamber at the Department of Botany, University of Dar es Salaam at a temperature range of 28 to 30°C. Light emitting diodes (LEDs) supplying 4.5 Klux light intensity at the surface of the vessels with a photoperiod of 12/12 h light/dark cycle were used as a source of light. Aerators fixed on the air pump were used to supply air in the cultures. On harvest, some of the spirulina concentrates was kept fresh in the refrigerator for further analysis and part of spirulina concentrates were oven dried at 60°C overnight. The dry sample was ground to make powder and then stored in the refrigerator.

Preparation of spirulina extracts

Fresh and dry biomass of spirulina (0.5 g) from each LCMA and Zarrouk media were placed in the conical flasks and then soaked in 100 ml of 95% ethanol. The sample was continuously stirred to ensure complete extraction. The extracts were centrifuged for 10 min then filtered using Whatman No. 1 filter paper. Ethanol was evaporated from the supernatant in a rotary evaporator (Gmbh & Co.KG, Germany) under reduced pressure at 40°C. Extraction was repeated three times until the desired concentration of extract was obtained. The obtained extracts were stored in a refrigerator at 4°C until further analyses.

Yield of extracts

The yield of ethanolic extracts of spirulina samples was calculated based on the following equation:

$$\text{Yield (\%)} = (W_2 \times 100) / W_1$$

Where W_1 is a weight of spirulina before evaporation/extraction and W_2 is the weight of extract after evaporation.

Determination of antioxidants levels and activity in spirulina extracts

Total phenolic compounds were estimated by Folin-Ciocalteu calorimetric method adapted from Pal et al. (2010). In brief, 1 ml of ethanolic extract was mixed with 1 ml of Folin-Ciocalteu's reagent

Table 1. Chemical composition of Zarrouk and LCMA culture media.

Component	Concentration (g/L)	
	Zarrouk	LCMA
NaHCO ₃	18	10
NaCl	1	1
MgSO ₄ ·7H ₂ O	0.2	-
NaEDTA	0.08	-
CaCl ₂ ·2H ₂ O	0.04	-
NaNO ₃	2.5	-
K ₂ SO ₄	1	-
K ₂ HPO ₄	0.5	-
FeSO ₄ ·7H ₂ O	0.01	-
NPK10-20-20 complex	-	0.5
Micronutrient	1 ml	1 ml
Distilled water	1 L	-
Boiled, cool tap water	-	1L (0.024 g/L N, 0.001 g/L P, 0.005 g/L K)

Micronutrients composition (g/L): H₃BO₃, 2.86; MnCl₂·4H₂O, 1.81; ZnSO₄·4H₂O, 0.222; Na₂MoO₄, 0.0177; CuSO₄·5H₂O, 0.08.

(Sigma-Aldrich, St. Louis USA) and incubated at room temperature. After 3 min, 1 ml of 7.5% (w/v) sodium carbonate was added to the mixture. The reaction was kept in the dark for 2 h and the absorbance was read at 725 nm using UV-spectrophotometer (Jenway 6305, UK). Standard solution of gallic acid was used to obtain a standard curve and total phenols were expressed as milligram of Gallic Acid Equivalent per gram of extract (mg GAE/g). Analysis for each extract was done in quadruplicate and the results were expressed as mean ± standard deviations (SD).

Total flavonoids concentration was determined by calorimetric assay according to Bonvehí et al. (2001). One milliliter of spirulina extract was diluted with 4.3 ml of 80% ethanol containing 0.1 ml 10% aluminium nitrate and 0.1 ml of aqueous potassium acetate (1 M). The mixture was left for 40 min at room temperature and then absorbance was read at 415 nm. Total flavonoids content expressed as Rutin Equivalents milligrams per gram of extract (RE mg/g) was calculated using rutin as standard.

The carotenoids contents, namely, β-carotene and lycopene were determined according to the method described by Barros et al. (2007). Only the dried samples of spirulina were used, as the method requires dry samples. Briefly, 100 mg of extract was shaken vigorously with 10 ml acetone-hexane for 1 min then filtered through Whatman No. 1 filter paper. The absorbance of filtrate was measured at 453, 505 and 663 nm and the contents of β-carotene and lycopene were calculated according to the following equations:

$$\text{Lycopene (mg/100 mg)} = -0.0458A_{663} + 0.372A_{505} - 0.0806A_{453}$$

$$\beta\text{-carotene (mg/100 mg)} = 0.216A_{663} - 0.304A_{505} + 0.452A_{453}$$

The mean value of three assays was used to estimate contents of carotenoids and results were presented as mg/g of extract.

DPPH radical scavenging assay

The ability of spirulina extracts to scavenge the stable radical 2,2-diphenyl-picrylhydrazyl (DPPH) was assessed based on the modified method of Batool et al. (2010). Briefly, a series of ethanolic extracts (0.01, 0.1, 0.2, 0.3, 0.4 and 0.5 mg/ml) were prepared. A measure of 500 µl of extract was mixed with 1 ml solution of DPPH

(0.125 µM in 95% ethanol). The mixture was shaken vigorously and incubated in dark room for 30 min and the absorbance was measured at 515 nm using UV-spectrophotometer. The DPPH radical scavenging activity (RSA) of spirulina extracts was calculated as percent of DPPH inhibition using the following equation:

$$\% \text{RSA} = ((A_{\text{DPPH}} - \text{AE}) / A_{\text{DPPH}}) \times 100$$

where A_{DPPH} is the absorbance of DPPH solution and AE is the absorbance of extract containing DPPH.

The percentage of DPPH RSA was plotted against spirulina extract concentrations (mg/ml) to determine the amount of extract necessary to inhibit (scavenge) initial concentration of DPPH radical by 50% (EC₅₀). The lower EC₅₀ value indicated higher scavenging activity of an extract.

Ferrous ions chelating assay

The ability of spirulina extracts to chelate Fe (II) ions was determined according to the method described by Pal et al. (2010). In short, different concentrations (0.01, 0.1, 0.2, 0.3, 0.4 and 0.5 mg/ml) of spirulina extracts and ethylene diamine tetra acetic acid (EDTA) were prepared. On start of experiment, 0.1ml of 2 mM FeCl₂ was added to the reaction mixture containing 10 mg of 0.1 M Tris-HCl and 0.2 ml 5 mM ferrozine and the aqueous ethanolic extracts. The reaction was left for 10 min at room temperature then the absorbance was measured at 562 nm against a blank in UV-spectrophotometer. The effective concentration (EC₅₀) of each extract at which 50% of ferrous ions were chelated, was obtained by interpolation from linear regression graph. EDTA was used as standard chelator.

Statistical analyses

All measurements were carried out in at least three replicates and the results presented as mean ± standard deviation. Data of the extract yields and antioxidants were analyzed by using

Table 2. The yields, contents of antioxidants, EC₅₀ values of DPPH scavenging and Fe²⁺ chelation in the Spirulina extracts.

Sample	Yield (%)	Phenolics (mg GAE/g) ^b	Flavonoids (RE mg/g) ^b	β-carotene (mg/100 mg) ^a	Lycopene (mg/100 mg) ^a	EC ₅₀ DPPH scavenging (mg/ml)	EC ₅₀ Fe ²⁺ chelation (mg/ml)
ZM fresh	6.98±0.00	137.65±4.39	5.10±0.12	NA	NA	0.3	0.078
LCMA fresh	6.83±0.01	151.45±0.70	8.30±0.141	NA	NA	0.202	0.068
ZM Dry	16.21±0.32	292.17±5.50	11.25±0.5	0.17±0.00	1.06±0.01	0.26	0.014
LCMA Dry	21.63±0.04	409.28±28.78	13.25±0.5	0.89±0.00	1.28±0.02	0.11	0.001

^aValues are the mean ± SD (n = 3); ^bvalues are mean ± SD (n = 4). ZM: Extracts from Zarrouk medium; LCMA: extracts from LCMA medium. NA: not applicable (the determination was done for dry samples only).

Paleontological Statistical programme (PAST ver. 2.17, Natural History Museum, University of Oslo, Norway). A two-sample *t*-test was used to see if there is significant difference in the yields, total phenolics, flavonoids and carotenoids between the two culture media. The differences between means at 5% (*P*-values less than 0.05) were considered significant.

RESULTS AND DISCUSSION

Yield of spirulina extracts

The extract yields of spirulina samples are shown in Table 2. The highest (21.63 ± 0.04%) and lowest (6.83 ± 0.01) yields for LCMA medium were recorded in dry and fresh extracts, respectively. In the fresh extracts, standard culture medium (Zarrouk) recorded significantly higher yield than LCMA (*p* < 0.001, *t* = 22.958). Similarly, there was significant difference (*p* < 0.0001, *t* = -28.709) among dried sample extracts with LCMA's extracts recording higher yield (21.63 ± 0.04%) as compared to Zarrouk's extracts (16.21 ± 0.32%). The highest yield recorded in LCMA suggests that LCMA is the best medium for the yield of spirulina biomass. The yields of the current study are higher than previously reported by Shalaby and Shanab (2013).

Antioxidant contents

Table 2 also shows the antioxidant contents in spirulina extracts. The total phenolics among other antioxidants were the most abundant. All the extracts analyzed were found to have significant amount of total phenols, flavonoids and carotenoids, which are evidence for protection of human body and other spirulina consumers against oxidative damage. With regard to total phenolics, it was shown that the dried sample extracts for spirulina grown in LCMA medium contained notably higher levels (409.28 ± 28.78 mg GAE/g) as compared to other extracts. There was significant variation in phenolics between dry (*p* = 0.0002, *t* = -7.8769) and fresh extracts (*p* = 0.0013, *t* = -5.6934). Differences in phenolic contents among extracts may be caused by several factors as

previous reported (El-Baky et al., 2009; Tarko et al., 2012; Salamatullah, 2014; Ismaiel et al., 2016). For instance, Tarko et al. (2012) stated that the composition of growth media used for cultivating the selected species of *Arthrospira* influenced the synthesis of bioactive components and antioxidant properties. However, Wu et al. (2013) reported that organisms may produce phenols as a defensive mechanism against disease and other stress especially when nutrient is depleted. The authors also recorded higher phenolic contents in the jujube tree planted in natural unfertilized area and they linked the observations with limitation of nutrient resources. Studies on *Spirulina* species (Salamatullah, 2014; Ismaiel et al., 2016), associated pH rise in the culture media to the increased production of phenols so as to alleviate the oxidative stress induced by the rising pH. In the current study, pH level (not reported here) was higher in LCMA medium in few days before harvest, and it was lower for Zarrouk medium; this might have influenced the variation in total phenols.

The current results on phenols are incomparable to previous reports due to the differences in methods and extraction solvent used. For instance, the study by Machu et al. (2015), water extract of *A. platensis* recorded the highest level of 43.2 mg/g GAE, which is lower than that obtained in this study. Another study by Shalaby and Shanab (2013), on *Spirulina platensis*, recorded phenolics of 282.76 g/100 mg, which are more or less similar to the current study (for Zarrouk's phenolics). However, Bhattacharya and Shivaprakash (2005) reported that higher phenolic contents as compared to the current study whereby *Spirulina laxissima* was found to contain the highest intracellular phenolics (4.46 g/100 mg), while *S. platensis* contained the highest extracellular phenolics (0.3 g/100 mg).

For the total flavonoids, there was significant difference in total flavonoids between dried (*p* = 0.0013, *t* = -5.6569) and fresh extracts (*p* < 0.0001, *t* = -35.054). The dried spirulina extracts from LCMA medium contained higher flavonoids than the dried extract of Zarrouk medium and the fresh extracts. In general, all extracts were found to have lower total flavonoids than the total phenols implying that large part of polyphenol compounds in

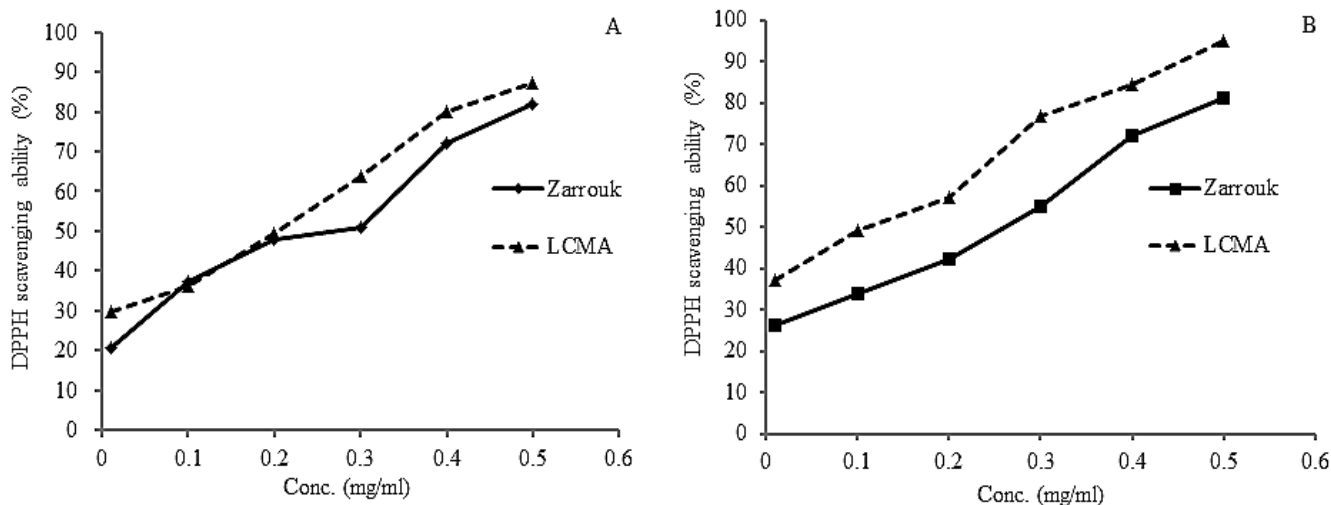


Figure 1. Scavenging activity (%) of spirulina ethanolic extracts on DPPH by fresh spirulina extracts (A) and dry extracts (B).

spirulina is phenolic substance instead of flavonoids. Other studies working on different sources of flavonoids (El-Baky et al., 2009; Salamatullah, 2014) also recorded lower amount of the flavonoids as compared to total phenols. As it is for phenolic compounds, the amount of flavonoid is also affected by the growth media (El-Baky et al., 2009).

Carotenoids are important groups of pigments found in some plants and algae. They possess strong antioxidant properties (Rao and Agarwal, 2000; Pal et al., 2010) and thought to be an anti-cancer agents (Gutiérrez-Salmeán et al., 2015). Beta carotene is a fat soluble pigment and is known as precursor of vitamin A in mammals (Pal et al., 2010), it bio-transforms into vitamin A once absorbed (Gutiérrez-Salmeán et al., 2015). In the current study, the content of β -carotene was higher in spirulina extract cultured in LCMA medium as compared to Zarrouk medium (Table 2). The reason for such variation may be due to the differences in the composition and amount of nutrients used for preparing the culture media. The investigations by Tarko et al. (2012) revealed that β -carotene contents of different strains of spirulina was strongly influenced by the growth medium whereby the standard Zarrouk medium was superior over the low-cost medium. In the current study, although the values of β -carotene are lower than reported by Tarko et al. (2012) but the LCMA recorded higher content than Zarrouk medium. Moreover, the β -carotene contents reported in the current study are much lower than previously reported (Bhattacharya and Shivaprakash, 2005; Belay, 2008; Gutiérrez-Salmeán et al., 2015) for other species of spirulina. Contrary, the content of β -carotene for *S. laxissima* reported by Bhattacharya and Shivaprakash (2005) was much lower than those in the current study. Earlier study demonstrated that the β -carotene is the most fluctuating pigment among other carotenoids, and

the variation can even be more than 40 times, that is, from 10 to 400 mg/100 g (Tarko et al., 2012).

For lycopene content, this study is the first found lycopene in spirulina (*A. fusiformis*). Spirulina extracts cultured in LCMA medium was found to have significantly ($p < 0.0001$, $t = -20.863$) higher lycopene contents than that of Zarrouk. The amount of lycopene reported here are slightly higher than that present in edible mushrooms (Barros et al., 2007; Robaszkievicz et al., 2010). Variations in lycopene contents are associated with variety of factors such as climatic conditions (Wawrzyniak et al., 2005) geographic location, fertilizer used and plant variety (Bhumsaidon and Chamchong, 2016).

Radical scavenging activity (RSA) using DPPH

The DPPH is a stable free radical, which possesses a characteristic absorption at 515 nm, the absorption decreases regularly upon exposure to radical-scavenging species. A lower absorbance indicates high radical scavenging activity of an extract (Barros et al., 2007). DPPH was selected to evaluate the antioxidant activities of spirulina extracts because it is the most effective and standard method for assessing the radical scavenging activity of a particular extract (Amarowicz et al., 2004; Maisuthisakul et al., 2007). Figure 1 shows the DPPH RSA of both fresh and dry spirulina extracts cultured in Zarrouk and LCMA media. The dry extract from LCMA exhibited strongest scavenging activity than all other extracts thereby quenching 95% of the DPPH radicals at the concentration of 0.5 mg/ml. The scavenging activities of other extracts at 0.5 mg/ml were 87, 81 and 72% for fresh spirulina in LCMA, dry and fresh spirulina in Zarrouk media, respectively. Though the extracts showed good scavenging activities, there was no significant variation

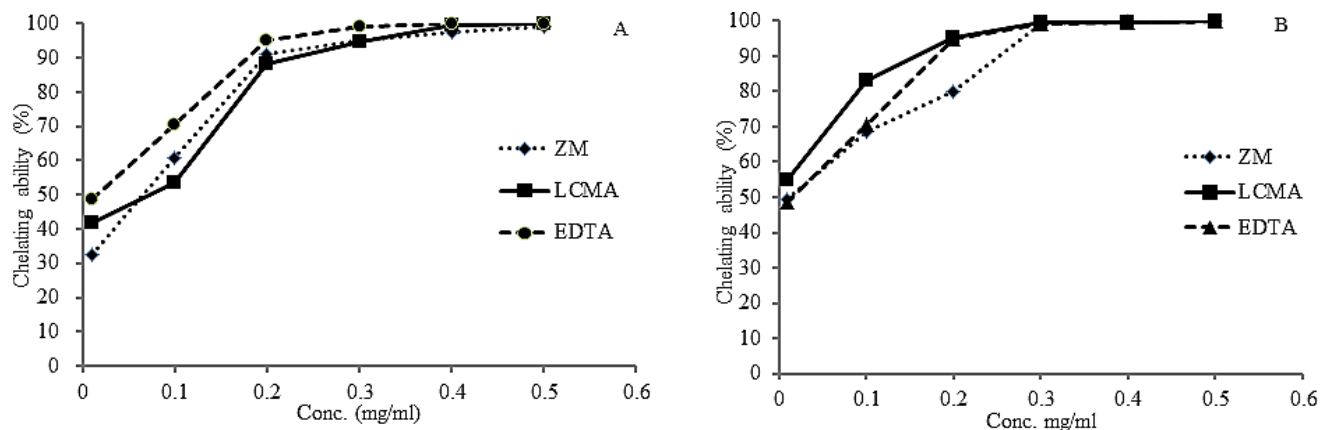


Figure 2. Ferrous (II) ion chelating ability of fresh spirulina extracts (A) and dry extracts (B). ZM: Zarrouk medium.

for both fresh and dried sample extracts ($p > 0.05$). Comparing with previous studies, spirulina extracts of the current study show stronger scavenging activity than reported by Chopra and Bishnoi (2008) for *A. platensis*. The result for dry extract cultured in LCMA is comparable to Miranda et al. (1998) who also reported a 95% antioxidant activity at 0.5 mg for *A. maxima* methanolic extract. The effective concentration (EC_{50}) usually corresponds to antioxidant activity of the particular extract (Maisuthisakul et al., 2007), the lower the EC_{50} value the powerful the antioxidant activity. The lowest EC_{50} (0.11 mg/ml) was recorded in dry spirulina extracts cultured in LCMA, which also contained high total phenols, flavonoids and carotenoids (Table 2). The EC_{50} values for extracts of fresh spirulina from LCMA, dry and fresh spirulina from Zarrouk media were 0.202, 0.26 and 0.3 mg/ml, respectively. However, the extracts with high contents of antioxidants did not always have lower EC_{50} value as it was evident for dry spirulina extract from Zarrouk medium which had higher contents of phenols and flavonoids but weak antioxidant activity, that is, higher EC_{50} value compared to fresh extract from LCMA. The reason for lower scavenging activity of the extracts with high antioxidants could be related to the types of phenols and flavonoids present in the extracts (Salamatullah, 2014). Differences in the strength of scavenging abilities between spirulina extracts of the current study might be due to differences in the compositions of culture media. For instance, Hussein et al. (2015) worked on antioxidants in the wild and domesticated mushroom extracts, reported that nutrients in the substrate had influence on the scavenging ability of the mushrooms. They suggested that higher scavenging ability of the domesticated mushroom was caused by high nutrient in the substrate (Hussein et al., 2015).

Ferrous ions chelating ability of spirulina extracts

The chelating power of extracts increased with increasing

concentrations (Figure 2). At 0.5 mg/ml concentration of fresh extracts of spirulina and standard chelator (Ethylene diamine tetra acetic acid, EDTA), chelating ability was in the order: EDTA (99.91%) > LCMA (99.87%) > Zarrouk medium (99.25%). The standard metal chelator showed an outstanding chelating capacity ($EC_{50} = 0.016$ mg/ml) compared to extracts from LCMA ($EC_{50} = 0.068$ mg/ml) and Zarrouk medium ($EC_{50} = 0.072$ mg/ml). For the dry samples, spirulina extracts showed powerful chelating power than the synthetic metal chelator (Figure 2B). Of the spirulina extracts, dried extract from LCMA medium revealed strong chelating power whereby it chelated 55% of the ferrous ions at lowest concentration (0.01 mg/ml). The EC_{50} value for dry extract of spirulina grown in LCMA was less than 0.01 mg/ml as the 50% of ions were chelated before attaining the initial concentration. The EC_{50} values for extract cultured in Zarrouk medium was 0.014 mg/ml while that of standard metal chelator (EDTA) was 0.016 mg/ml. In general, it was observed that both fresh and dried spirulina extracts possess strong chelating activity against the ferrous ions as they chelated more than 50% of the ions at lower concentration of 0.1 mg/ml. However, there was no significant difference in chelating power between the dry samples as well as between the fresh sample extracts ($p > 0.05$).

Conclusion

The current study revealed that both fresh and dried extracts of *A. fusiformis* are rich source of antioxidants with a substantial amount of total phenols, flavonoids, β -carotene, and lycopene. Thus, consumption of spirulina as diet or feed/food additive is important for health. Spirulina extracts grown in the newly formulated low-cost medium resulted in higher antioxidants and stronger scavenging activity as compared to the standard culture medium. This implies that the production of natural antioxidants from spirulina can be maximized through the

use of cost-saving culture medium, the LCMA. However, it was noted that sometimes the extracts with higher antioxidants do not always express powerful antioxidant activity. This calls for further studies to assess the types of phenols and flavonoids in spirulina extracts.

CONFLICT OF INTERESTS

The authors have not declared any conflict of interests.

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Full Length Research Paper

Nutritional potential of underutilized gum arabic tree seeds (*acacia nilotica*) and locust bean seeds (*Parkia biglobosa*)

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***Acacia nilotica* seed (ANS) and *Parkia biglobosa* seed (PBS) are underutilised legume found to have health benefits and functional properties. This study determined nutrient composition of *A. nilotica* and *P. biglobosa* seeds. ANS and PBS were collected and processed properly for chemical analysis. The proximate, minerals, vitamins, essential amino-acids, and antinutrient composition were analyzed to ascertain nutritional attributes and its potential in promoting dietary diversity. The raw and fermented *A. nilotica* seeds contained in g/100 g, protein (12.88 to 15.38), fat (3.29 to 4.91), ash (5.24 to 6.84), dietary fibre (1.98 to 2.66) and available carbohydrate (69.63 to 71.73), while the FPB contained in g/100 g, protein (18.30), fat (9.20), ash (8.69), dietary fibre (2.61), and available carbohydrate (56.27). The fermented *A. nilotica* (FAN) seed contained all the nine (9) essential amino acids. The raw and fermented ANS contained in mg/100 g, iron (9.67 to 12.23), zinc (0.69 to 1.13), calcium (0.17 to 0.22), sodium (0.14-0.21) while the FPB seed contained 14.86, 1.59, 0.25, and 0.24, respectively. FAN and FPB contained in µg/100 g, vitamin A (148.79 and 197.81), vitamin E (15.90 and 24.69) and vitamin K (1.41 and 1.63), respectively. The levels of antinutrient factors in all the samples were not significant. Fermented *A. nilotica* seed contained adequate level of some micronutrients and essential amino acids. Consumption of the seed should therefore be promoted.**

Key words: Nutrient composition, dietary diversity, underutilized legumes, essential amino acid, functional properties, proximate composition, antinutritional factors.

INTRODUCTION

Globally, the nutritional value of legumes is gaining considerable interest due to the demand for healthy and nutritious foods. Presently, attention towards underutilized

legumes is increasing to find new alternate protein sources to meet the ever increasing demand for vegetable protein (Pugalenthi et al., 2005). In developing

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countries, underutilized legumes households particularly, during drought, famine and dry season (Magbagbeola et al., 2010). Legumes are the life-savers for millions of resource poor people where ensuring food and nutritional security is one of the significant problems, particularly in traditional subsistence farming systems (Haq, 2002). Deficiencies of micronutrients are a major global health problem. It is estimated that 1.2 billion people in the world do not have enough food to meet their daily requirements and a further 2 billion people in the world today are estimated to be deficient in key vitamins and minerals, particularly vitamin A, iodine, iron and zinc (FAO, 2012).

Dietary diversification is a proven cost-effective strategy to ameliorate malnutrition. The loss of dietary diversity has many implications for the nutrition and health of rural communities including loss of income generation and decreased consumption of diverse foods. The multiple benefits of preservation and promotion of indigenous foods range from a collateral benefit on biodiversity and environmental sustainability to improving micronutrient intakes (Bharucha and Pretty, 2010). In many communities in south west Nigeria, edible fruits from wild plants are often taken as food or added to food as condiment to supplement important minerals and vitamins in human diets (Olujobi, 2012). In order to address protein malnutrition and food security issues in developing countries, there is need to emphasize the utilization of legumes as a low-cost dietary vegetable protein source.

Acacia nilotica is a species of *Acacia* (wattle) native to Africa and the Indian subcontinent. In India, it is called gum arabic tree, kikar, babul or Indian gum arabica tree and recognized worldwide as a multipurpose tree (National Academy of Sciences, 1980). *A. nilotica* has both nutritional and medicinal values based on the presence of numerous secondary metabolites and essential metals (Bwai et al., 2015). However, *A. nilotica* has not attracted much attention in Nigeria and it is an under-exploited legume that has many medicinal properties found mainly on the bark, stem, pods, gum, trunk, and leaves of the gum arabic tree (*A. nilotica*).

A. nilotica seeds can be fermented and serve as a seasoning like fermented *P. biglobosa* and also provide essential macro and micro nutrients when consumed. This seed has the potential to assume a more important role globally in the sustainable supply of diverse and nutritious food if given appropriate attention by agriculturists and nutritionists. Hence, this study determined the nutrient composition of *A. nilotica* seeds and compares them with *P. biglobosa*, to find out if *A. nilotica* seeds are nutritious and safe for consumption.

MATERIALS AND METHODS

Sample collection

A. nilotica seed pods were collected from Odo-Owa in Ijero Local

Government Area, Ekiti State, Nigeria. Freshly fermented *P. biglobosa* seed samples were purchased from Oja Oba market; a local main market situated in Ado-Ekiti, Ekiti State, Nigeria.

Sample preparation and analysis

The seed pods were separated from the fruit. Foreign materials and dirt were removed from the seed pods by hand picking and washing. The seedpods were boiled in a pressure pot for 5 to 6 h. When it was well boiled, the seed pods were dehulled and the whole cotyledons were separated from the pods. The dehulled cotyledons were washed properly with distilled water, and drained appropriately. They were further cooked in distilled water for 1 h, drained, and allowed to cool, then separated in two places. One half of the raw cooked seeds was dried in its solid form in an air oven at 65°C for 20 h, blended and packaged properly for chemical analysis, while the other half was spread in calabash lined with banana leaves (*Musa acuminata*) and was well covered with more banana leaves. Fermentation process was carried out for 4 days. The fermented samples were oven dried at 65°C for 20 h, blended (using Marlex; Excella mixer grinder with 3 stainless steel jars, made in India. Reg. trademark no: 277985) and finally packaged for chemical analysis. The fermented *P. biglobosa* seed samples was also dried in an air oven at 65°C for 20 h, blended (using Marlex; Excella mixer grinder with 3 stainless steel jars, made in India. Reg. trademark no: 277985) and packaged for chemical analysis. The sample preparation was done in the dietetics kitchen, Afe Babalola University Ado-Ekiti.

Proximate composition and amino acid determination

Moisture content of raw and fermented *A. nilotica* seed, and fermented *P. biglobosa* were analyzed by drying method, crude fat by Soxhlet extraction method and crude protein by semimicro-Kjeldhal method (AOAC, 1996). Crude ash was determined by incinerating in a muffle furnace at 550°C (AOAC, 1996). Dietary fibre was determined by enzymatic gravimetric method – Prosky (AOAC, 1995). Available carbohydrate value was calculated as the difference between 100 and the sum of the percentages of water, protein, crude fat, ash and dietary fibre.

The samples were mixed in a laboratory blender, hydrolyzed at 150°C for about 90 min and the solution was used for the determination of amino acids by the modified Waters 'Pictotag' system described by Bidlingmeyer et al. (1984).

Determination of minerals

Potassium and sodium content of the samples were determined by digesting the ash of the samples with perchloric acid and nitric acid, and then taking the readings on Jenway digital flame photometer/spectronic20 (Bonire et al., 1990). Calcium, magnesium, iron, zinc, copper and selenium content of the samples were determined from the digested ash of the samples spectrophotometrically by using Buck 200 atomic absorption spectrophotometer (Buck Scientific, Norwalk) (Essien et al., 1992) and compared with absorption of standards of these minerals.

Vitamin analysis

Vitamin A determination (AOAC Method 960.5 & 974.29, 2005)

Vitamin A was determined through ultraviolet absorption measurement at 328 nm after extraction with chloroform.

Calibration curve of vitamin A acetate was made and sample vitamin A concentration estimated as microgram (μg) of vitamin A acetate.

Thiamine (Vitamin B₁) determination (Woollard and Indyk, 2002)

Thiamine content of the samples was determined by weighing 1 g of each sample into 100 ml volumetric flask and adding 50 ml of 0.1 M H₂SO₄ and boiled in a boiling water bath with frequent shaking for 30 min. Five milliliters of 2.5 M sodium acetate solution was added and flask set in cold water to cool contents below 50°C. The flask was stoppered and kept at 45 to 50°C for 2 h and thereafter made up to 100 ml mark. The mixture was filtered through a No. 42 Whatman filter paper, discarding the first 10 ml. 10 ml was pipetted from remaining filtrate into a 50 ml volumetric flask and 5 ml of acid potassium chloride solution was added with thorough shaking. Standard thiamine solutions were prepared and treated same way. The absorbance of the sample as well as that of the standards was read on a fluorescent UV spectrophotometer (Cecil A20 Model) at a wavelength of 285 nm.

Riboflavin (Vitamin B₂) determination (AOAC Official Method 981.15, 1995)

One gram of each sample was weighed into a 250 ml volumetric flask, 5 ml of 1 M HCl was added, followed by the addition of 5 ml of dichloroethene. The mixture was shaken and 90 ml of de-ionized water was added. The whole mixture was thoroughly shaken and heated on a steam bath for 30 min to extract all the riboflavin. The mixture was then cooled and made up to volume with de-ionized water. It was then filtered, discarding the first 20 ml of the aliquot. 2 ml of the filtrate obtained was pipetted into another 250 ml volumetric flask and made up to mark with de-ionized water. Sample was read on the fluorescent spectrophotometer at a wavelength of 460 nm. Standard solutions of riboflavin were prepared and readings taken at 460 nm, and the sample riboflavin obtained through calculation.

Niacin (Vitamin B₃) determination (AOAC Methods 944.13, 985.34, 2000)

Five grams of blended sample was extracted with 100 ml of distilled water. Five millilitres of this solution was drawn into 100 ml volumetric flask and made up to the mark with distilled water. Standard solutions of niacin were prepared and absorbance of sample and standard solutions were measured at a wavelength of 385 nm on a spectrophotometer and niacin concentration of the sample estimated.

Determination of Vitamin B₆, B₉ and B₁₂ (Antakli et al., 2015)

Vitamin B₆, B₉ and B₁₂ were determined with simple ion-pair RP-HPLC validated method using a C18 column with two different detectors, photodiode array detector (UV-DAD) and fluorescence detector (FLD).

Ascorbic acid determination (AOAC official Method 967.21, 2005)

Ascorbic acid in the sample was determined by titrating its aqueous

extract with solution of 2,6-dichlorophenol-indophenol dye to a faint pink end point.

Tocopherol (Vitamin E) determination

One gram of sample was weighed into a 250 ml conical flask fitted with a reflux condenser wrapped in aluminium foil, and refluxed with 10 ml of absolute ethanol and 20 ml of 1 M ethanolic sulphuric acid for 45 min. The resultant solution was cooled for 5 min, followed by addition of 50 ml of distilled water and then transferred into a separating funnel covered with aluminium foils. The unsaponifiable matter in the mixture was extracted with 5 x 50 ml diethyl ether. The combined extract was washed free of acid and dried over anhydrous sodium sulphate. The extract was later evaporated at a low temperature and the residue obtained was immediately dissolved in 10 ml absolute alcohol. Aliquots of solutions of the sample and standard were transferred to a 20 ml volumetric flask. 5 ml absolute ethanol was added, followed by a careful addition of 1 ml concentrated HNO₃ and placed on a water bath at 90°C

Determination of Vitamins D and K

Vitamins D and K were determined by methods described by AOAC Official Method 2016.05 and 2015.09, respectively

Antinutrient determination

The phytate content was determined using the method adopted by Mohammed et al. (1986). Oxalates were determined by the official method of analysis (AOAC, 1990) and tannins determined as described by Bainbridge et al. (1996). The quantification of saponin levels was done using afrosimetric method (Koziol, 1990) while the trypsin inhibitory activity was determined on casein and comparing the absorbance with that of trypsin standard solutions read at 280 nm (Makkar and Becker, 1996).

Data analysis

Statistical Package for Social Sciences (SPSS) for Windows, version 15.0 (SPSS Inc. Chicago, IL, USA) was used for data analysis. The means and standard error of mean were calculated for all values. *A. nilotica* and *P. biglobosa* samples were compared using analysis of variance (ANOVA). Differences between samples were determined using Fisher's least significant difference (LSD) test. $P < 0.05$ was considered significant.

RESULTS AND DISCUSSION

Proximate composition

Proximate composition of *Acacia* and *P. biglobosa* seeds were compared in Table 1. The three samples were low in moisture content, moderate in dietary fibre and crude fat and high in crude protein, ash and carbohydrate content. The difference between the samples is significant ($p < 0.05$) with locust bean seed (*P. biglobosa*) having a more favoured proximate values. Proximate composition is an important criterion used to determine the nutritional values and quality of food (Qayyum et al., 2012). The low moisture content of the fermented *A. nilotica* seed is desirable because low moisture content

Table 1. Proximate composition of gum arabic tree seeds (*A. nilotica*) and locust bean seeds (*P. biglobosa*) (g/100 g).

Parameters	Raw <i>Acacia nilotica</i>	Fermented <i>Acacia nilotica</i>	Fermented <i>Parkia biglobosa</i>
Moisture	3.08±0.02 ^a	2.39±0.03 ^b	4.92±0.02 ^c
Crude protein ^a	12.88±0.05 ^a	15.38±0.06 ^b	18.30±0.01 ^c
Crude fat ^a	4.91±0.01 ^a	3.29±0.02 ^b	9.20±0.02 ^c
Fibre ^a	2.66±0.01 ^a	1.98±0.02 ^b	2.61±0.01 ^a
Ash ^a	6.84±0.07 ^a	5.24±0.02 ^b	8.69±0.01 ^c
Carbohydrate ^a	69.63±0.08 ^a	71.73±0.02 ^b	56.27±0.07 ^c

Values are mean ±SEM, n =3. Value with different superscripts on the same row are significantly different ($P<0.05$). ^aValues are on dry weight basis.

Table 2. Essential amino acid (%) in gum arabic tree seeds (*A. nilotica*), and locust bean seeds (*P. biglobosa*).

Parameters	Fermented <i>A. nilotica</i>	Fermented <i>P. biglobosa</i>
Arginine	2.99±0.02 ^a	4.18±0.01 ^b
Isoleucine	3.31±0.02 ^a	4.91±0.02 ^b
Leucine	6.54±0.01 ^a	9.41±0.02 ^b
Histidine	2.01±0.03 ^a	2.40±0.02 ^b
Methionine	1.51±0.02 ^a	1.79±0.01 ^b
Lysine	5.37±0.01 ^a	8.09±0.01 ^b
Threonine	4.91±0.01 ^a	7.05±0.01 ^b
Phenylalanine	3.11±0.01 ^a	4.69±0.01 ^b
Valine	6.29±0.02 ^a	8.30±0.01 ^b

Values are mean ±SEM, n =2. Value with different superscripts on the same row are significantly different ($P<0.05$). Values are on dry weight basis

discourages the growth of bacteria and mold, and enhances stability and shelf life.

Fermented *P. biglobosa* seed has the highest protein value (18.30 g/100 g) as compared to the fermented acacia seed (15.38 g/100 g) and raw acacia seed (12.88 g/100 g). Fermentation increase protein content. Fagbemi et al. (2005) reported protein increase in fermented fluted pumpkin (*Telfairia occidentalis*) seeds for production of 'Ogiri ugu' also, Enujiugha (2003), Omafuvbe et al. (2004) and Pelig-Ba (2009) observed similar trends from their studies. The increase in amino acid content with fermentation time is especially important from the nutritional point of view as it would increase digestibility and absorption. Dietary proteins usually play an important role in natural synthesis and maintenance of body tissues, enzymes and hormones as well as other substances that are required for healthy functioning (Hayat et al., 2014).

Slight decrease was observed in ash content of the fermented *A. nilotica* seed as compared to the raw seeds. Loss in ash contents may be due to leaching of soluble inorganic salts into the processing water during the

fermentation period (Effiong and Umoren, 2011) or the fermenting microflora used for their metabolism (Oladunmoye, 2007).

Essential amino acids profile

The essential amino acids (EAA) profile in fermented *A. nilotica* seed and locust bean is shown in Table 2. Fermented *P. biglobosa* was significantly higher in all the EAAs examined when compared with fermented *A. nilotica* seed ($p<0.05$). The result also reveals that EAAs account for more than half of amino acids found in fermented *P. biglobosa*, while that of fermented *A. nilotica* seed was about 36%.

All legumes are rich in lysine, phenylalanine and arginine but limiting in sulphur containing amino acids (methionine and cystine). The most concentrated essential amino acid in the fermented *P. biglobosa* seed and fermented *A. nilotica* seed studied was leucine. A study carried out by Ogunbusola et al. (2010) showed similar observation for *Lagenaria siceraria* seed flour and

Table 3. Minerals composition gum arabic tree seeds (*A. nilotica*) and locust bean seeds (*P. biglobosa*) (mg/100 g).

Parameter	Raw <i>A. nilotica</i>	Fermented <i>A. nilotica</i>	Fermented <i>P. biglobosa</i>
Sodium	0.14±0.00 ^a	0.21±0.00 ^b	0.24±0.00 ^c
Potassium	0.33±0.00 ^a	0.59±0.00 ^b	0.73±0.00 ^c
Magnesium	0.21±0.00 ^a	0.24±0.00 ^b	0.27±0.00 ^c
Calcium	0.17±0.00 ^a	0.22±0.00 ^b	0.25±0.00 ^c
Zinc	0.69±0.02 ^a	1.13±0.02 ^b	1.59±0.02 ^c
Iron	9.67±0.01 ^a	12.23±0.03 ^b	14.86±0.01 ^c
Copper	0.16±0.01 ^a	0.39±0.02 ^b	0.67±0.02 ^c
Selenium	0.12±0.00 ^a	0.28±0.02 ^b	0.51±0.01 ^c

Values are mean ±SEM, n =3. Value with different superscripts on the same row are significantly different ($P < 0.05$). Values are on dry weight basis.

its protein fractions. The lowest value of essential amino acid amongst the fermented *P. biglobosa* seed sample and fermented *A. nilotica* seed sample was methionine (a sulphur containing amino acid).

Mineral composition

Minerals composition of raw gum arabic tree seeds (*A. nilotica*), fermented gum arabic tree seeds and fermented locust bean seeds (*P. biglobosa*) is shown in Table 3. The result reveals that the three samples were low in sodium, potassium, magnesium and calcium, moderate in zinc, copper and selenium and were good sources of iron.

There was a general increase in the mineral content of the fermented *A. nilotica* seeds. This may be due to the processing method used: fermentation has been found to increase the nutritional content of foods. The increase in these minerals may also be due to the contribution from fermenting organisms as stated by Ugbogu et al. (2014). The low sodium/potassium ratio of the fermented *A. nilotica* seed samples is an indication that it may reduce the incidence of hypertension as it would not induce high blood pressure, which is the major cause of cardiovascular diseases (Du et al., 2014), while the calcium content in the fermented *A. nilotica* seed samples implies that it could aid in preventing bone diseases such as rickets, osteoporosis and osteomalacia. Notably, calcium also enhances the effective use of iron in the system (Adeyeye, 2013).

The iron content in fermented *A. nilotica* seeds was noteworthy as diets in many developing countries are deficient in iron (Bressani, 2002). It is required for a number of biological functions, including proper functioning of the immune system, electron transfer reactions, gene regulation, cell growth and differentiation as well as binding and transport of oxygen (Siddiqui et

al., 2014). The values from the fermented seed samples are higher than that of Ogunyinka et al. (2017) who carried out a study on fermented *P. biglobosa*. Magnesium is the only predominant mineral amongst the three seed samples. This may signify that legumes are good sources of magnesium. Although, the value from a study carried out by Ogunyinka et al. (2017) is higher than the magnesium content of these seeds. This may be due to the time the study was carried out.

Magnesium is required for the action of more than 300 enzymes in the body, where it participates in several significant physiological functions in the maintenance of good health and glucose homeostasis. In addition, it has been identified to play a significant role in the release of insulin and the maintenance of pancreatic β -cells. Its deficiency has been implicated in insulin resistance and carbohydrate intolerance as well as diabetic complications and dyslipidemia (Piero et al., 2012; Akhuemokhan et al., 2013; Khan and Awan, 2014).

Zinc was appreciably detected in the fermented *A. nilotica* seed samples. Zinc is known to enhance immune health and play a crucial role in antioxidant defense in type 2 diabetic patients where it enhance reduction and neutralization of free radicals and acts as a cofactor of super oxide dismutase (SOD), by modulating glutathione metabolism and metallothionein expression. Its deficiency has been implicated in a number of metabolic abnormalities such as impaired glucose tolerance, decreased pancreatic insulin content as well as insulin degradation (Piero et al., 2012; Cruz et al., 2015). Copper is a trace element that is required in little amount by the body for the formation of hemoglobin, it is also a part of energy metabolism enzyme that was found in notable amounts in the fermented *A. nilotica* seed samples. Finally, selenium was found to be low in the fermented *A. nilotica*. Selenium is also a trace element that is an important antioxidant which works with vitamin E to promote immune system response.

Table 4. Vitamin composition of *A. nilotica* and *P. biglobosa* seeds (mg/100 g).

Parameter	Raw <i>A. nilotica</i>	Fermented <i>A. nilotica</i>	Fermented <i>P. biglobosa</i>
Vitamin A	89.06±0.01 ^a	148.79±0.02 ^b	197.81±0.02 ^c
Vitamin B ₁	0.33±0.01 ^a	0.58±0.02 ^b	0.81±0.01 ^c
Vitamin B ₂	0.02±0.00 ^a	0.03±0.00 ^b	0.05±0.00 ^c
Vitamin B ₃	1.07±0.01 ^a	1.23±0.01 ^b	1.38±0.01 ^c
Vitamin B ₆	1.13±0.01 ^a	1.38±0.02 ^b	1.57±0.01 ^c
Vitamin B ₉	0.01±0.00 ^a	0.02±0.00 ^b	0.03±0.00 ^c
Vitamin B ₁₂	0.21±0.01 ^a	0.31±0.02 ^b	0.47±0.01 ^c
Vitamin C	0.17±0.01 ^a	0.35±0.01 ^b	0.59±0.01 ^c
Vitamin D	0.09±0.01 ^a	0.15±0.01 ^b	0.31±0.01 ^c
Vitamin E	9.62±0.01 ^a	15.90±0.02 ^b	24.69±0.01 ^c
Vitamin K	1.16±0.01 ^a	1.41±0.01 ^b	1.63±0.01 ^c

Values are mean ±SEM, n =3. Value with different superscripts on the same row are significantly different ($P<0.05$). Values are on dry weight basis.

Table 5. Anti-nutritional factors of *A. nilotica* and *P. biglobosa* seeds (%).

Parameter	Raw <i>Acacia nilotica</i>	Fermented <i>Acacia nilotica</i>	Fermented <i>Parkia biglobosa</i>
Tannin	0.00±0.00 ^a	0.00±0.00 ^a	0.00±0.00 ^a
Saponin	0.05±0.00 ^a	0.07±0.00 ^b	0.01±0.00 ^c
Phytate	0.00±0.00 ^a	0.01±0.00 ^b	0.01±0.00 ^b
Oxalate	0.00±0.00 ^a	0.00±0.00 ^a	0.01±0.00 ^c
Trypsin	0.18±0.01 ^a	0.36±0.02 ^b	0.81±0.00 ^c

Values are mean ±SEM, n =3. Value with different superscripts on the same row are significantly different ($P<0.05$). Values are on dry weight basis.

Vitamin composition

Vitamin composition of the samples of raw *A. nilotica* seed, fermented seed and fermented *P. biglobosa* seed is presented in Table 4. The three samples were low in water soluble vitamins such as B₁, B₂, B₁₂ and vitamin C, but contained reasonable levels of B₃ and B₆. The three samples were good sources of fat soluble vitamins except vitamin D. Vitamins A and E are important antioxidants that are usually responsible for protecting the body tissue from damage caused by free radicals that can harm cells, tissues, and organs in the human body. The lower values of vitamins B₁, B₂, B₉, B₁₂, and C in the seed sample could probably be due to leaching during processing (Nelson and Cox, 2000). The water soluble vitamin, vitamin B₃ is an important nutrient that can be used to improve cholesterol levels, reduce atherosclerosis, and treat high levels of triglycerides and LDL cholesterol (<http://www.naturalremedies.org/water-soluble-vitamins/>). Vitamin B₆ (pyridoxine) is responsible for the buildup of some neurotransmitters in the brain that contribute to elevated mood and decreased depression. For this reason, it has been used in the treatment of

some mood disorders as a supplement. The recommended daily allowance of vitamin B₆ is 1.3 to 1.7 mg; depending on age. Vitamin B₆ is also helpful in reducing cardiovascular disease in the body and maintaining good heart health. Finally, it is an essential component of the absorption of Vitamin B₁₂ by the body and the two vitamins work closely together for health maintenance (<http://www.naturalremedies.org/water-soluble-vitamins/>).

Antinutritional factors

The result of antinutritional factors presented in Table 5 shows that raw, fermented *A. nilotica* seed and fermented *P. biglobosa* seeds were free of tannin, very low in saponin, phytate and oxalate, and slightly high in trypsin inhibitors. Many traditional methods of food preparation such as fermentation, cooking and malting increase the nutritive quality of plant foods by reducing certain anti-nutrients such as phytic acid, polyphenols and oxalic acids (Sarkar and Nout, 2014). The values detected are at a safe level and poses no danger in diets. This is in

agreement with what was reported previously for African locust beans (*P. biglobosa*) (Ijarotimi and Keshinro, 2012). One of the major factors that contribute to a lower nutritional value of legumes and cereals is the presence of naturally-occurring anti-nutritional factors (ANFs, for example phytate, tannins and enzyme inhibitors) that indirectly decrease the bioavailability status of minerals. However, these ANFs can be reduced through different processing methods, such as heating, enzymes application, soaking, sprouting, irradiation, fermentation, mechanical methods such as dehulling and milling, or even by means of other techniques such as high pressure processing, microwave heating and extrusion (Nooshin et al., 2017).

Conclusion

In conclusion, this study has shown the nutritional composition of *A. nilotica* and *P. biglobosa* seed. *A. nilotica* seed is not known and consume unlike *P. biglobosa* seed. This study has demonstrated that *A. nilotica* seed is rich in macro- and micro-nutrients and low in antinutritional factors. Further research using animal study should be conducted to help determine bioavailability of nutrients and safety of consumption.

CONFLICT OF INTERESTS

The authors declare that there is no conflict of interest.

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Full Length Research Paper

Physicochemical, microbiological and sensory characteristics of cashew milk formulated yoghurt

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Cashew yoghurt was formulated using cashew kernel milk. The physicochemical, microbiological and organoleptic assessment of the resulting product was examined and compared with commercial yoghurts. Results indicated that crude protein of 16.8% for cashew milk yoghurt compared favorably with 14.82% of commercial yoghurt. Caloric value of cashew yoghurt (133.06 kJ) was higher than that obtained for commercial yoghurts (112.01 kJ). Microbiological study revealed a total count of 1×10^6 CFU/g, which was attributed to the culture used in the fermentation process. The cashew milk yoghurt was generally accepted by the panelist as there were no differences in the sensory characteristics examined; however, there was a difference at $p > 0.05$ for mouth feel for the sample studied. It can be concluded that cashew milk serves as a potential recipe for acceptable yoghurt formulation.

Key words: Cashew kernel, yoghurt, sensory, protein, organoleptic.

INTRODUCTION

Cashew plant, *Anacardium occidentale* L. is native to Brazil. The plant is widely grown in the continents of Asia, Europe as well as Africa. In most of the African countries, especially Nigeria, the tree is grown mainly for its pseudo apple and the nuts, which are mainly consumed raw, that is, unprocessed (Ohler, 1979; Nambiar et al., 1980). Cashew is a climacteric crop, and harvesting of the nuts often takes place between January and April or May. When the cashew apples are ripe, they fall down from the tree, and the nuts are collected after detaching from the rotten apples, sun-dried on concrete floor for two days to

attain moisture content of 12%, and this is to avoid spoilage. Nigeria's annual cashew nut production from 466,000 MT in the year 2000 to 836,500 MT in year 2012 representing 45% of cashew nuts produced in African (FAOSTAT, 2013).

The fruit of cashew tree consists mainly of the cashew nut, an embryo shaped shelled nut and the false fruit cashew apple. The composition of the shelled nut consists of 20 to 25% kernel, 20 to 25% cashew nut shell liquid (CNSL), 20% testa and 48 to 55% shell. The cashew nuts are majorly sold as export crop and few of

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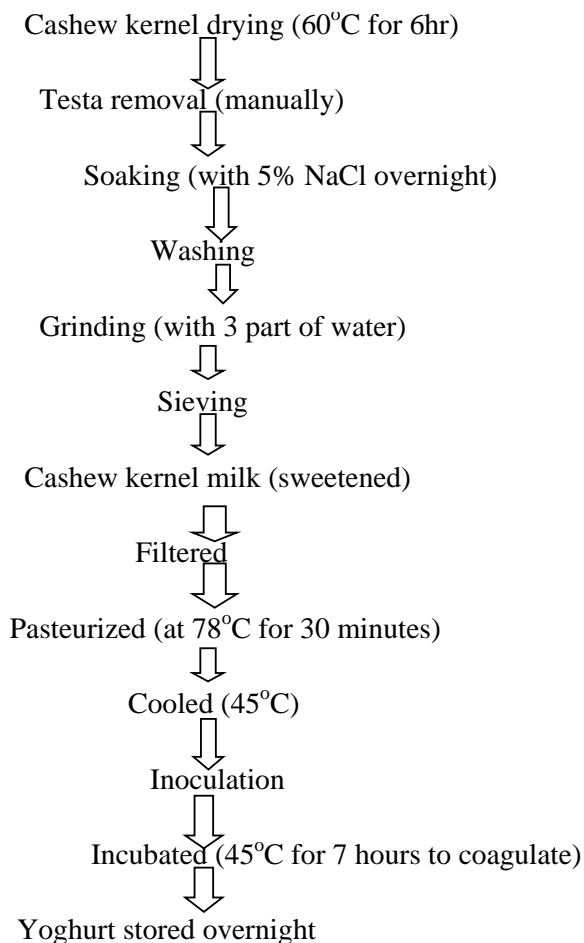


Figure 1. Flow chart for cashew kernel formulated yoghurt processing.

the nuts are roasted, flavored, and sold for local consumption in different packaging materials. Embedded in the honeycomb pericarp of the nut is a viscous brown liquid- cashew nut shell liquid, CNSL, that is, phenolic in nature. It is a by-product of the cashew nut industry. The components of this liquid have been found to be Anacardic acid (71.7%), cardol (18.7%), cardanol (4.7%), novel phenol (2.7%) and an unknown minor ingredient (2.2%) (Tyman and Morris, 1967). The liquid has been identified as an important industrial raw material (Ghatge and Maldar, 1981). For example, CNSL has been recognized as a good source of unsaturated phenol, an excellent monomer for polymer production. It is a good natural alternative to petrochemically derived phenol.

The sensory and nutritional qualities of cashew kernels cannot be overemphasized. For example, cashew kernels are excellent source of proteins (20 to 24 g/100 g), carbohydrates (23 to 25 g/100 g) and fats (40 to 57 g/100 g) (Nascimento et al., 2010; Ogunwolu et al., 2009; Yang, 2009). Moreover, studies have shown that cashew

kernels have beneficial effects on health, particularly on chronic diseases such as hypertension and obesity, coronary heart disease and diabetes. However, the high content of unsaturated fatty acids of most nut kernels is one of the most determinant factor against cardiovascular disease and obesity (Mexis and Kontominas, 2009; Oliete et al., 2008; Yang, 2009; Yang et al., 2009; Aswal et al, 2012; Horáčková et al., 2015).

Formulation of value added products from cashew kernel had received much attention from various workers over the years (Sofu and Ekinci, 2007; do Espirito Santo et al., 2010; Chen et al., 2010). There is dearth of information on the use of cashew kernel milk in yoghurt formulation, and this work seeks to be addressed.

MATERIALS AND METHODS

Cashew nut kernel

Cashew nuts were obtained from Ochaja Station of Cocoa Research Institute of Nigeria. The nuts were sun dried until ready for use. The proximate composition of the kernel was carried out according to standard methods of analysis (AOAC, 1990).

Cashew kernel milk preparation

Cashew kernels were removed from nuts using manual cashew kernel breaker. The kernels were dried at oven temperature of 60°C for 6 h for easy removal of testa. Cashew kernels (250 g) were soaked in 5% NaCl solution overnight. The soaked kernel was cleaned and ground to a smooth paste. The cashew kernel paste was diluted with 3 parts of water, sieved and the supernatant was sweetened with 16 g of sugar, 0.6 g of vanilla flavor, bottled and pasteurized at 78°C for 30 min and refrigerated.

Formulation of cashew milk yoghurt

To obtain cashew kernel yoghurt, the produced milk was inoculated with *Streptococcus thermophilus* and incubated at 45°C for 7 h to coagulate. The resulting yoghurt (Figure 1) was then stored overnight in a bath of cold water.

Physicochemical analyses of formulated yoghurt

The pH of the formulated product was measured using a standard pH meter (pH meter kent EK 7020). Titratable acidity (TA %) was determined by titration of sample against 0.1 N NaOH (Speck, 1984). Moisture content, crude fat, protein and total ash content of the formulated yoghurt were determined according to the AOAC (1990) and Pearson (1976). The refractometric method as described by Akinsaye (1998) was used to determine the sugar content of the yoghurt. In a typical experiment, 20 ml of yoghurt was mixed with 10 ml of lead acetate (10%) in a beaker and filtered through Whatman’s filter paper (No. 4) into 100-ml volumetric flask. Two spoonful of sodium hydrogen carbonate was added to the filtrate to precipitate excess lead and then filtered. The filtrate was used for the refractometric determination of the sugar content using Abbe 60 Refractometer. The viscosity was measured using viscometer.

Table 1. Chemical composition of cashew kernel.

Parameters	Value (dry wt)
Water	6.92%
Protein	21.52%
Ether extract	47.00%
Carbohydrate	29.30%
Crude fiber	1.13%
Ash	3.3.2%
Food energy	2602.9 kJ/100 g

Values are percentage mean of triplicate determinations.

Microbiological analysis

Formulated cashew kernels yoghurt samples were analyzed for total *Escherichia coli*, *Staphylococcus aureus*, *Salmonella* spp. and *Clostridium* spp., according to the American Public Health Association (Downes and Ito, 2001) guidelines.

Sensory analyses

A 9-point hedonic scale was used to measure the sensory qualities, that is, colour, aroma, mouth feel/texture, taste and overall acceptability of the product (Larmand, 1977). Ten member in-trained panelists that are used to yoghurt taste were employed to ascertain or detect any difference between the cashew milk yoghurt and the locally purchased commercialized products. These samples were coded differently and served to the panelists with a glass of water and were instructed to rinse their mouth in between the tasting period. The scale of preference ranges from 9 representing like extremely to 1 representing dislike extremely.

Statistical analysis

The results obtained were subjected to analysis of variance (ANOVA). Mean comparisons were carried out between the two yoghurt samples by Turkeys multiple range test and by Statistical Programs for Social Sciences (SPSS, 1992).

RESULTS AND DISCUSSION

Physicochemical analysis

Table 1 presents results of the proximate composition of cashew nut flour. The mean moisture value of cashew kernel was $6.92 \pm 0.2\%$, dry weight. This is a little below the value for some legumes ranging between 7 and 11%. The low moisture content helps to reduce microbiological deterioration. The ash content obtained for this study indicates that it is low and may not be suitable for animal feed. Results from the study shows that cashew kernel contains appreciable amount of crude fat (47%) and this makes it a potential and economically viable vegetable oil. Fat is important in diet because it promotes fat soluble

vitamin absorption (Bogert et al., 1994; Champagne et al, 2010; Lee and Lucey, 2010; Shahnawaz et al., 2013). It is a high energy nutrient and does not add to the bulk of the diet.

Proximate composition of yoghurt

Table 2 shows the nutritional composition of commercially sold yoghurt as compared to that of cashew kernel milk. A pH value of 4.10 was obtained for commercial yoghurt as compared to 4.20 for cashew kernel yoghurt. This obtained result could be a reflection of the souring activity of lactic and this also explain the high titratable acidity (%), 0.5 to 0.55 obtained. Reed (1982) noted in his work that good quality yoghurt should have pH of 4.15 and TA (% lactic acid) of 0.5. The values obtained in this work are similar to these stated values. From the result, the calorific value of the cashew kernel yoghurt (133.06 KJ) was higher than that of the commercial yoghurt (112.01KJ). This means that cashew yoghurt can be taken as both proteinous and energy food.

The crude ash content of commercial samples, 0.76%, is lower than that of cashew kernel yoghurt (0.84%). Crude protein (%) results showed that cashew kernel yoghurt of 16.88 was higher than the commercial yoghurt (14.82). The use of cashew kernel in yoghurt production improves the crude protein value remarkably. Similarly, the crude fat content (%) ranges from 36.01 of commercialized yoghurt to 40.32 of cashew kernel yoghurt. Egan et al. (1981), Reeds (1982) and Terna and Musa (1998) stated in their work that commercial yoghurt should have the following minimum proximate compositions, which are 3.5% for protein, 3.25 fat and 87.7% moisture. This result is higher than recommended values obtained by these authors. The coagulation time of the cashew kernel yoghurt sample was 6 h. It has been reported that the duration of fermentation at 40 to 45°C is 3 to 6 h (Ebing and Rutgers, 1996) and 3 to 5 h at 45°C (Kosikowski, 1982). The fermentation time for this

Table 2. Proximate composition of cashew yoghurt.

Analyses	Yoghurt commercial	Cashew yoghurt
pH	4.1	4.20
TA (%)	0.50	0.55
Moisture content (%)	69.0	64.52
Crude protein (%)	14.8	16.88
Crude fat (%)	36.09	40.32
Total soluble solids	59.20	56.22
Ash (%)	0.76	0.84
Specific gravity	1.0212	1.0320
Viscosity (sec.)	205.7	203.4
Caloric value (KJ)	112.01	133.06
Dry matter (%)	14.47	14.65

Table 3. Mean values of bacteriological analyses of cashew kernel yoghurt and commercialized yoghurt.

Identity	Test	Result
Cashew kernel yoghurt	Total count (37°C)	1 x 10 ⁶ CfU/g*
	<i>E. coli/coliforms</i>	Nil
	<i>Staphylococcus aureus</i>	Nil
	<i>Salmonella</i> spp.	Nil
	<i>Clostridium</i> spp.	Nil
Commercialized yoghurt	Total count	1 x 10 ⁶ cfu/g
	<i>E. coli</i>	Nil
	<i>Staphylococcus aureus</i>	Nil
	<i>Salmonella</i> spp.	Nil
	<i>Clostridium</i> spp.	Nil

*Cfu/g– Colony forming unit per gram.

experiment (7 h) is in agreement with the above authors. Lactose is the fermentable sugar generally preferred by lactic acid bacteria despite non addition to the formulated milk, the product was able to exhibit the souring ability of the lactics.

Milk pasteurization (75°C) was done to modify milk protein to enhance proper viscosity and gelatinization of the product. This is in agreement with the work of Reed (1982). This also resulted in uniformity and smoothness in body texture of the yoghurt samples as indicated in Table 2.

Microbiological assay

The bacteriological quality of cashew kernel yoghurt samples and that of the commercial yoghurt samples is depicted in Table 3. Results revealed that total count of the samples is 1 x 10⁶ cfu/g, while specific count of *E.*

coli, *S. aureus*, *Salmonella* spp. and *Clostridium* spp. were negative for both samples. This showed that the total counts as shown in table were because of the cultures used to ferment the yoghurt, thereby resulting in the production of lactic acid, which is lethal to some organisms. Lactic acid bacteria also produce hydrogen peroxide, diacetyl and bacteriosis as antimicrobial substances which create hostile environment for food-borne pathogen and spoilage organisms, and therefore are able to suppress the multiplication of pathogenic and putrefying bacteria. It is believed that pasteurization temperature of 75°C for 30 min would have effectively destroyed any microbes present in the milk samples coupled with the low pH values (Chumchuere and Robinson, 1999; Rodriguez et al., 2010). In addition, the traditional starter cultures used in cashew kernel yoghurt production contain substantial quantities of beta-D-galactosidase enzyme and consumption of cashew kernel yoghurt may assist in alleviating the symptoms of

Table 4. Mean comparison of sensory evaluation scores of cashew kernel yoghurt and locally commercialized yoghurt.

Yoghurt	Aroma	Colour	Taste	Mouth feel	Overall acceptability
Cashew kernel yoghurt	8.1NS	7.2NS	7.4NS	8.8a	8.0NS
Commercial yoghurt 1	7.8	7.0	7.4	7.0b	7.8
Commercial yoghurt 2	7.3	7.0	7.0	7.5ab	8.0
Skimmed milk yoghurt	8.0	6.6	6.9	8.2a	7.5

Letters with the same column are significantly different ($P < 0.05$); NS- not significant.

lactose maladsorption. The cashew kernel yoghurt was therefore considered safe microbiologically.

Sensory characteristics

The results of the organoleptic appraisal performed on all the yoghurts are shown in Table 4. The results indicates that panelist find it difficult to differentiate them from one another. The products have milky color with semi-solid texture and the taste was slightly sour. However, there was a significant difference at $P > 0.05$ for mouth feel. The overall acceptability showed that the yoghurt samples were acceptable but does not show any significant difference at $P < 0.05$.

Conclusion

This study shows that yoghurt can be produced from a vegetable source especially for the vegetarians. This indicates that it is possible to produce good quality yoghurt that is well acceptable for human consumption using cashew kernel. It was also shown that there were no differences in the taste and flavor of yoghurts studied and this makes the product more enjoyable and pleasurable. Thus, cashew kernel yoghurt due to its content and viscosity can delay gastric emptying and can be of an immense benefit for people who are lactose intolerant.

CONFLICT OF INTERESTS

The authors have not declared any conflict of interests.

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