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# **Short Communication**

# Potentialization of antioxidant activity of *Piper*umbellata (Piperaceae) leaves after their metabolism in Heraclides brasiliensis larvae (Lepidoptera: Papilionidae)

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*Piper umbellata* leaves accumulate the compound, 4-nerolidylcatechol, which is responsible for the antioxidant power of the plant. Surprisingly, in this study we observed a significant increase in the antioxidant activity of *P. umbellata* leaves after their metabolism on *Heraclides brasiliensis* larvae. The total phenolic concentrations of the extracts from the leaves and fecal matter (metabolized leaves) measured using the Folin-Ciocalteu reagent, were 14.78 and 27.13 mg GAE/g, respectively. The antioxidant activity, based on 2,2-azinobis 3-ethylbenzothiozoline-6-sulfonic acid (ABTS), was EC<sub>50</sub> of 120.10 and 62.50 ug/ml for the leaves and fecal material, respectively. These values were correlated with the total phenolic content.

**Key words:** Piper umbellata, Heraclides brasiliensis, antioxidant, metabolism.

### INTRODUCTION

Piper umbellata (L.) Miq. (Piperaceae) [syn. Heckeria umbellata (L.) Kunth., Pothomorphe umbellata L., Piper hilarianum Stend.], known popularly as "caapeba" in Brazil and "cowfoot" in English speaking countries, has been reported to be a potent antioxidant (Kijjoa et al., 1980). This antioxidant activity of P. umbellata is attributed to its major constituent, 4-nerolidylcatechol (1), detected in its root and leaf tissues (Pinto et al., 2010; Ropke et al., 2003; Noriega et al., 2008a; b). Others minor compounds (N-p-coumaroyl tyramine, piperumbellactam A, B and C) isolated of P. umbellata branches showed antioxidant activity (Tabopda et al., 2008). The compound 1 is biotransformed to E-2,3dihydro-3-(3,4-dihydroxyphenyl)farnesoic metabolism of the leaves by Heraclides brasiliensis, a host insect (Figure 1) (Ramos et al., 2012). Whereas alterations in the chemical profile of plant tissues can

drastically affect their various biological proprieties, the present work aimed to evaluate the antioxidant properties of *P. umbellata* leaves after their metabolism in *H. brasiliensis*.

# **MATERIALS AND METHODS**

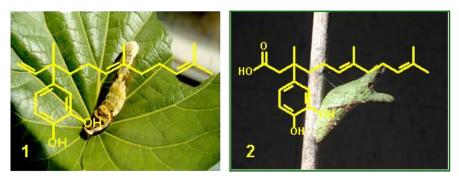
# Chemicals

Folin-Ciocalteu's phenol reagent, potassium persulfate and 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid (Trolox) were obtained from Sigma-Aldrich (Steinheim, Germany), and 2,2-azinobis 3-ethylbenzothiozoline-6-sulfonic acid (ABTS) was supplied by Fluka Chemie (Buchs, Switzerland).

# Botanical and entomological materials

Leaves of *P. umbellata* were collected in the city of Itabuna, Bahia, Brazil, in November 2011. A voucher specimen was deposited in the herbarium of the Institute of Biosciences of the University of São Paulo, under code KATO-671. This plant was authenticated by Dr. Elsie F. Guimarães of the Jardim Botânico Research Institute in Rio de Janeiro, RJ, Brazil. Larvae of *H. brasiliensis* were collected

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**Figure 1.** Chemical structures of compounds **1** (4-nerolidylcatechol) and **2** (*E*-2,3-dihydro-3-(3,4-dihydroxyphenyl)farnesoic), larvae and pupae of *H. brasiliensis* in *P. umbellata*.

Table 1. Results of antioxidant activity of leaves and fecal extracts.

Extract	Total phenol mg EAG/g ± DP	ABTS EC <sub>50</sub> (ug/ml) ± DP
Leaves	14.78 ± 3.29	120.10 ± 1.70
Fecal	27.13 ± 4.53	62.50 ± 0.60
Trolox	-	$1.95 \pm 0.02$

mg EAG/g = mg of gallic acid per g of extract;  $EC_{50}$  = effective concentrations.

collected from *P. umbellata* in November 2011 in Itabuna and identified by Dr. Sérgio Antônio Vanin of the Zoology Department of the Biosciences Institute of the University of São Paulo. A voucher specimen (CSR 006) was deposited at the Zoology Museum of the University of São Paulo.

Dried leaves of *P. umbellata* (10 g, 40°C for 48 h) were milled and subjected to ethanol extraction (120 m) twice, and then concentrated in a vacuum to yield 0.91 g of crude extract. Five individuals of *H. brasiliensis* were collected on *P. umbellata* leaves and were maintained in cages for a month under a diet consisting of *P. umbellata* leaves, with feces collected daily. Dried feces (2 g, 40°C for 48 h) were milled and subjected to two methanol extractions (100 ml), which after concentration in a vacuum yielded 0.25 g of crude extract.

### **LC-MS Analysis**

HPLC analysis of extracts and pure compounds was performed with a Shimadzu chromatograph model SCL-10A with UV–VIS detector, using a Supelco reversed phase column. Elution was carried out in a gradient mode starting with acetonitrile:water (4:6) for 10 min, rising to (9:1) in 30 min and maintained for to 35 min. The flow rate was 1 mL/min; injection volume 20  $\mu L$ ; UV scan, 200-400 nm, and all chromatograms were obtained at  $\lambda$  max = 214 nm.

# **Antioxidant activity**

Significant levels of antioxidant activity of the leaf and fecal extracts were found using 2,2'-azino-bis(3-ethylbenzthiazoline-6-sulfonic acid) (ABTS\*+) radical scavenging assays (ABTS\*+) scavenging method). The 1,1-diphenyl-2-picrylhydrazyl radical (DPPH\*) scavenging activity of the isolated compounds was measured as reported in the literature (Slinkard and Singleton, 1977) with some modifications. Briefly, ABTS\*+ was generated by reacting ABTS solution (7 mM) with potassium persulphate (2.45 mM, final concentration) for 12 - 16 h in the dark at room temperature. Then, the ABTS\*+ solution

was diluted with ethanol to obtain absorbance of 0.70 ( $\pm$  0.02) at 734 nm and further equilibrated at 30°C. The fecal and leaf extracts were diluted in ethanol at a concentration of 1 mg/ml, and the ABTS\*\* solution was added to both extracts to obtain concentrations of 1.0 to 100.0 µg/ml. The absorbance of the reaction mixture was measured at 734 nm after reaction at room temperature for 10 min. Trolox was used as positive control. The capability of scavenging the ABTS+\* radical was calculated using the following equation:

ABTS\*\* scavenging effect (%) = [( $A_0 - A_1 / A_0$ ) × 100] where  $A_0$  is the initial concentration of ABTS\*\* and  $A_1$  is the absorbance of the remaining concentration of ABTS\*\* in the presence of the sample.

### **Total phenolic content**

The total phenolic content of the fecal and leaf extracts was determined using the Folin-Ciocalteu reagent according to the method of Slinkard and Singleton (Duh et al., 1999), modified by the use of gallic acid as a standard phenolic compound. Appropriate amounts of extracts (500  $\mu L;$  50  $\mu g/mL)$  were diluted in a volumetric flask with distilled water (3 ml). The Folin-Ciocalteu reagent (100  $\mu L)$  was added and the contents of the flask were thoroughly mixed. After 3 min, Na<sub>2</sub>CO<sub>3</sub> (15%, 300  $\mu L)$  was added and the mixture was completed with distilled water (5 ml) and allowed to stand for 2 h in an ultrasonic bath. The absorbance was measured at 760 nm in a spectrophotometer. The total amount of phenolic compounds was determined in micrograms of gallic acid equivalents, using the equation obtained from the standard gallic acid graph.

# RESULTS AND DISCUSSION

The chemical profiles of the leaf and fecal extracts obtained

In summary, the potentialization of the antioxidant activity of leaves of *P. umbellata* after metabolism in *H. brasiliensis* can be associated with the biotransformation of compound **1** into **2** as well as the increase of the total phenolic content of the fecal extract. This study type contributes to the discovery of new natural bioactive compounds because it describes, for the first time as a novelty, the modification of a natural source (*P. umbellata* leaves) with potent antioxidant activity by insect.

# **ACKNOWLEDGEMENTS**

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