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Full Length Research Paper

Hepatoprotective and antioxidant activities of *Pterocarpus santalinoides* methanol leaf extract

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This study evaluated the hepatoprotective and antioxidant activities (AA) of Pterocarpus santalinoides methanol leaf extract (PSMLE) on carbon tetrachloride (CCI_a)-induced hepatotoxicity in albino rats. Thirty male albino rats randomly assigned into 6 groups (A – F) of 5 rats each were used for the in vivo study. Hepatotoxicity was induced in groups A – E using CCl₄. Group A served as negative control. Groups B, C and D were treated with 50, 250 and 500 mg/kg PSMLE, respectively. Group E was treated with 100 mg/kg Silymarin, while Group F served as normal control. Treatment was given orally twice daily for 15 days, after which markers of hepatotoxicity and oxidative stress were evaluated. The in vitro AA of PSMLE was also evaluated using 1, 1-diphenyl 2-picryl hydrazyl. Results showed that treatment with PSMLE at 250 and 500 mg/kg led to significantly (p<0.05) lower serum alanine aminotransferase and malondialdehyde, significantly (p<0.05) higher superoxide dismutase and glutathione peroxidase levels, while 250 mg/kg dose further led to significantly (p<0.05) lower serum aspartate aminotransferase and serum total bilirubin levels, and significantly (p<0.05) higher serum total protein and serum globulin levels. 500 mg/kg dose treatment additionally led to significantly (p<0.05) lower serum total cholesterol. Treatment at all doses led to significantly (p<0.05) lower liver weight and relative liver weights and significantly (p<0.05) higher catalase and total glutathione levels. The PSMLE exhibited significantly (p<0.05) higher AA at concentrations \geq 50 µg/ml in vitro. It was concluded that PSMLE was hepatoprotective and possesses significant antioxidant activity in vivo and in vitro.

Key words: Hepatotoxicity, oxidative stress, antioxidants, *Pterocarpus santalinoides* leaf extract, carbon tetrachloride.

INTRODUCTION

The liver is a vital organ for metabolism, excretion, clearance and transformation of chemicals in the body

(Singh et al., 2011). It is responsible for the detoxification of drugs and xenobiotics; thus, it is constantly and

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Author(s) agree that this article remain permanently open access under the terms of the <u>Creative Commons Attribution</u> <u>License 4.0 International License</u> variedly exposed to xenobiotics which may induce liver damage (Saukkonen et al., 2006). Most absorbed toxins and toxicants will first pass through liver, and the possible response elicited may range from inflammation to degeneration and/or neoplasia of the hepatocytes (Schiff and Schiff, 1987). Hepatotoxicity is a major health problem, and the manifestations vary from asymptomatic elevation of liver enzymes to fulminant liver failure (Saukkonen et al., 2006). Toxic liver damage is commonly oxidative stress mediated, and constitutes a large proportion of liver disorders/diseases; its occurrence has been steadily increasing over the years (Suk and Kim, 2012, Rehm et al., 2013; Nwokediuko et al., 2013).

Carbon tetrachloride (CCl₄) is a commonly used model chemical for the experimental induction of hepatotoxicity (Kim et al., 2010). It is metabolized to trichloromethyl (CCl₃) free radical which induces hepatotoxicity by causing peroxidative degradation in the adipose tissue, resulting in fatty infiltration of the hepatocytes (Boll et al., 2001). Following administration, CCl₄ is activated by cytochrome CYP_{2E1} and CYP_{2B1} to form CCI_3 radical which binds to cellular molecules such as nucleic acids, proteins and lipids, thereby impairing crucial cellular processes like lipid metabolism, with the potential outcome of fatty degeneration (Boll et al., 2001). The CCl₃ radical reacts with oxygen to form highly reactive species, the trichloromethylperoxy (CCl₃OO) radical, which initiates the chain reaction of lipid peroxidation culminating in destruction of polyunsaturated fatty acids (Boll et al., 2001). This causes alteration in permeability of the mitochondria, endoplasmic reticulum, and plasma membranes, resulting in the loss of cellular calcium, disruption of calcium homeostasis and damage/death of hepatocytes (Weber et al., 2003).

Oxidative stress is a state in which oxidation and oxidants exceed the antioxidant systems in the body leading to imbalance between the generation of reactive oxygen species (ROS) and the level of antioxidants in the biological system (Yoshikawa and Naito, 2002). It occurs when free radicals which are not neutralized by antioxidants go on to create more volatile free radicals and damage cell membranes, vessels, proteins, fats and DNA. Biological free radicals are highly unstable reactive molecules that have electrons available to react with various organ substrates such as DNA, proteins and lipids. Oxidative stress is known to be involved in the pathogenesis of a variety of diseases including atherosclerosis, hypertension, diabetes mellitus, ischemic diseases, liver diseases and malignancies (Yoshikawa and Naito, 2002), or may exacerbate their symptoms (Halliwell and Gutteridge, 1989; Valko et al., 2007).

Antioxidants are compounds that inhibit the oxidation of other compounds and prevent chemical damage caused by free radicals (Sies, 1997). Oxidation reactions in living organisms produce free radicals which can initiate chain reactions that may cause damage or death to cells. Antioxidants terminate these chain reactions by removing free radical intermediates and inhibiting other oxidation reactions (Sies, 1997; Valko et al., 2007). Insufficient levels of anti-oxidants or inhibition of the antioxidant enzymes in living organisms cause oxidative stress which may lead to injury and/or death of cells (Davies, 1995; Valko et al., 2007). Many of the natural antioxidants such as tannins, flavonoids and glycosides are very important in the prevention of diseases associated with oxidative stress (Yi-Fang et al., 2002; Aruoma, 2003).

Some plants such as Cussona barteri (leaves). Lannea vilutina (leaves), Sacoglotis gabonensis (stem bark), Trichilia roka (roots), Tinospora cordifolia (whole plant), Piptadeniastrum africanum (stem bark) and Gongronema latifolium (leaf) amongst others, have been reported to be rich sources of natural antioxidants that can protect against oxidative stress and thus play important role in the chemoprevention of diseases that have their etiology and pathophysiology in ROS (Ames et al., 1993; Atawodi, 2005; Karamalakova et al., 2018; Dlamini et al., 2019). There has been an increase in interest in the therapeutic potential of plants as antioxidants that may reduce free (Schuler. radical-induced tissue iniurv 1990: Karamalakova et al., 2018). A number of plants such as Ipomoea batatas (leaves), Allium cepa (leaves), Cnestus ferruginea (leaves stem and roots), Splenacentrum jollyanum (leaves and roots) and Voacanga africana (leaves) had been investigated in the search for novel antioxidants (Chu, 2000; Mantle et al., 2000; Koleva et al., 2002; Oke and Hamburger, 2002), while a lot more are still under investigation.

Pterocarpus santalinoides DC is an indigenous Nigerian plant in the family Papilionaceae (Keay, 1989). It is commonly known as "red sandal wood" in English language and "nturukpa" in Igbo language (Adetunji, 2007; Anowi et al., 2012). Leaves of *P. santalinoides* are used traditionally as vegetable for making soup, and also in folk medicine for the treatment of various ailments including heart and liver diseases (Adesina, 1982; Okwu and Ekeke, 2003). Leaves of P. santalinoides contain considerable amounts of tannins, flavonoids and glycosides which are known natural antioxidants (Anowi et al., 2012; Eze et al., 2012; Ihedioha et al, 2017; 2018). Previous studies by Ihedioha et al. (2017) suggested that the hepatoprotective properties of methanol leaf extract of Ρ. santalinoides in acetaminophen-induced hepatotoxicity may be attributed to its antioxidant phytochemical composition. Also, reports from earlier studies on the lipid lowering effects of aqueous leaf infusion of *P. santalinoides* in guinea pigs showed that it may be related to its antioxidant properties (Ihedioha et al., 2018). Based on the various traditional medicinal uses of P. santalinoides especially in the treatment of diseases in which oxidative stress is known to play critical role and the results of these earlier cited studies, the purpose of the present study was to evaluate the

hepatoprotective and antioxidant activities of methanol leaf extract of *Pterocarpus santalinoides* on CCl₄-induced sub-acute hepatotoxicity *in vivo* in albino rats, and *in vitro* using the DPPH assay method.

MATERIALS AND METHODS

Plant collection, identification and extract preparation

Fresh leaves of *Pterocarpus santalinoides* (Figure 1) used for the study were collected from Nsukka Local Government Area of Enugu State in November 2017. The plant was identified and authenticated by a plant taxonomist (Mr. A.O. Ozioko) at the Department of Plant Science and Biotechnology, University of Nigeria, Nsukka, a voucher specimen [UNH (University of Nigeria Herbarium) No. 02] was deposited at the University of Nigeria, Nsukka, herbarium. The leaves were dried under shade and pulverized. Five hundred grammes of the pulverized leaves were extracted with 80% methanol using the cold maceration extraction technique. The resulting extract was filtered with Whatman size 1 filter paper, concentrated to dryness with a Rotary Evaporator (Buchi, Switzerland), and referred to as *Pterocarpus santalinoides* methanol leaf extract (PSMLE).

Experimental animals

Thirty adult male albino rats (Rattus norvegicus) of 12 weeks of age, weighing between 200 and 250 g, were obtained from the Laboratory Animal House of the Department of Veterinary Physiology and Pharmacology, University of Nigeria, Nsukka, and used for the in vivo hepatotoxicity study. Fifteen adult female rats of 12 weeks of age weighing between 172- 190 g were also obtained from the same source and used for the acute toxicity study. The albino rats were housed in stainless steel cages in a fly proof Animal House at room temperature between 23 -29°C, and allowed 2 weeks to acclimatize before the commencement of the study. They were fed commercial rat pellets (Grand Cereals Nig. Ltd, Jos, Nigeria), composed of 13% crude protein, 8% fat, 15% crude fibre, 0.9% calcium, 0.35% phosphorus and 2600 Kcal/kg metabolizable energy, and clean drinking water ad libitum. They were cared for and handled humanely all through the study. Stipulated guidelines governing the use of animals for laboratory experiments were strictly adhered to (Zimmermann, 1983; Ward and Elsea, 1997). The protocol for the laboratory animal study was approved by the Faculty of Veterinary Medicine Institutional Animal Care and Use Committee, University of Nigeria, Nsukka, (Approval No: FVM-UNN-IACUC/2018/0814).

Acute toxicity study

The acute toxicity and median lethal dose (LD_{50}) of PSMLE were determined in albino rats following the OECD Acute Toxic Class method (OECD, 2001). Fifteen adult female albino rats, randomly assigned into 5 groups of 3 rats each (Groups 1, 2, 3, 4 and 5) were used for testing at 0, 50, 300, 2000 and 5000 mg/kg, respectively. The albino rats were fasted for 12 hours before the test commenced, but water was made available to them *ad libitum*. The extracts were each dissolved in 1 ml of distilled water and administered orally with a gastric tube. The rats were observed for 14 days for any sign of toxicity or mortality. Their body weights were

measured at intervals, and at the end of the 14 days of observation, they were humanely sacrificed and the weights of their vital organs (liver, kidney, spleen and heart) were measured (OECD, 2001).

Phytochemical analysis of PSMLE

Phytochemical analysis was done to determine the phytochemical constituents of PSMLE, following standard procedures as described by Trease and Evans (1996), Harborne (1998). One gramme (1 g) of PSMLE was dissolved in 100 ml of distilled water in a beaker. The solution was filtered with Whatmann no. 1 filter paper to obtain a clear filtrate which was used to test for the presence of tannins, flavonoids, alkaloids, saponins, phenols, carbohydrates, glycosides, starch, polyuronides, sterols and terpenes. High level presence of each phytochemical was scored +++, moderate levels were scored ++, low levels were scored + (Trease and Evans, 1996; Harborne, 1998).

Evaluation of the effects of PSMLE on blood levels of markers of hepatotoxicity, oxidative stress and antioxidant marker levels in albino rats given sub-acute toxic doses of CCI₄

The thirty male albino rats used for the hepatotoxicity and oxidative stress study were randomly assigned to 6 groups A - F of five rats each. Hepatotoxicity and oxidative stress were induced in rats in groups A – E using the CCl4 model (Robin et al., 2012; Singh et al., 2012). A mixture of 1 ml/kg CCl₄ in equal volume of olive oil (50% v/v) was injected intraperitoneally to groups A – E at 3 day intervals (days 0, 3, 6, 9 and 12) for 12 days. Group A was treated with 10 ml/kg distilled water as placebo and served as negative (untreated) control, Groups B, C and D were treated with 50, 250 and 500 mg/kg PSMLE, respectively. Group E was treated with 100 mg/kg Silymarin (a known hepatoprotective and antioxidant drug) as positive control, while Group F was also given 10 ml/kg distilled water as placebo and served as normal control (not given CCl₄). Treatment started 24 h post-initial CCl₄ administration and was done twice daily for 15 days. On day 15 post-initial CCl₄ administration, two milliliters of blood sample was collected from each rat and used immediately for evaluation of blood levels of enzyme markers of liver damage, oxidative stress and antioxidant markers, following standard procedures. Blood sample collection was done using the orbital technique (Bolliger and Everds, 2010). A portion of the blood (1.5 ml) was dispensed into a glass test tube and allowed to stand at room temperature for 45 min to clot; then centrifuged at 3000 revolutions per minute for ten minutes using a table centrifuge (Jenalab Medical, England), after which the serum was harvested and assayed immediately. The remaining 0.5 ml was dispensed into a heparinized sample bottle for use in the determination of glutathione peroxidase activity.

Evaluation of effects of PSMLE on blood levels of markers of hepatotoxicity

Serum alanine aminotransferase (ALT) and serum aspartate aminotransferase (AST) activities were evaluated following the Reitman and Frankel method (Colville, 2002), using the Quimica Clinica Applicada (QCA) serum ALT and AST test kits. The ALT in the serum sample and standards catalyzed the reaction of Lalanine and alpha-ketoglutaric acid to form pyruvic acid and Lglutamic acid, while the AST catalyzed the reaction of L-aspartic acid with alpha- ketoglutaric acid to form oxaloacetic acid and Lglutamic acid. These ketonic acids reacted with 2-4, dinitrophenyl hydrazine to form corresponding colored hydrazones, the optical



Figure 1. Leaves of Pterocarpus santalinoides.

density of which was measured and ALT/AST quantified at 505 nm wavelengths, using the Diatek[®] semi-automated blood biochemistry analyzer (Diatek Instruments, Wuxi, China).

The serum alkaline phosphatase (ALP) activity was quantified using the QCA alkaline phosphatase test kit, based on the phenolphthalein monophosphate method (Colville, 2002). The alkaline phosphatase in the serum and a standard (containing 30 IU/L alkaline phosphatase) hydrolyzed a colorless substrate of phenolphthalein monophosphate and gave rise to phosphoric acid and phenolphthalein which at alkaline pH turned into pink color. The optical density was measured and the alkaline phosphatase activity was quantified at 546 nm wavelength, using the semi-automated biochemistry analyzer (Diatek Instruments, Wuxi, China).

Serum total protein levels were determined using the QCA total protein test kit which was based on the direct Biuret method

(Lubran, 1978; Johnson, 2008). A reaction of the proteins in the serum samples and a standard (containing 5 g/dl of proteins), with copper ions in the Biuret reagent in an alkaline medium, resulted in the formation of a stable colored complex. The optical density of the colored complex was measured at 546 nm wavelength and quantified using the Diatek[®] semi-automated blood biochemistry analyzer (Diatek Instruments, Wuxi, China).

Assay of serum albumin was done using the QCA albumin test kit based on the bromocresol green method (Doumas and Peters, 1997; Johnson, 2008). This involved the reaction of the albumin in the serum samples and standard (containing 5 g/dl of albumin) with bromocresol reagent at acid pH to form a colored complex. The optical density of the colored complex was measured at 630 nm wavelength and the serum albumin was quantified using the Diatek[®] semi-automated blood biochemistry analyzer (Diatek Instruments, Wuxi, China). The globulin levels were calculated by subtracting the serum albumin levels from the total protein levels (Johnson, 2008).

The serum total cholesterol levels were determined using the QCA total cholesterol test kit, which is based on the enzymatic colorimetric method (Allain et al., 1974; Rifai et al., 2008). Total cholesterol in the serum samples and standard (containing 200 mg/dl of cholesterol) was enzymatically hydrolyzed by cholesterol esterase and further oxidized by cholesterol oxidase contained in the QCA total cholesterol working reagent. The reactions resulted to formation of a colored quinonic derivative. The optical density of the colored quinonic solution was measured at 505 nm wavelength and the total cholesterol was quantified using the Diatek[®] semi-automated blood biochemistry analyzer (Diatek Instruments, Wuxi, China).

The total bilirubin levels in the serum samples was assayed using the Randox[®] bilirubin test kit (Randox Laboratories Ltd, County Antrim, United Kingdom), which is based on the Jendrassik and Grof method (Doumas et al., 1973; Higgins et al., 2008). The serum samples were reacted with diazotized sulfanilic acid in the presence of caffeine to produce an azopigment. Their optical densities were measured at 578 nm and quantified a Chem5V3[®] Semi-automated Clinical Chemistry Analyzer, (Erba Diagnostics, Mannheim GmbH, Mannheim, Germany).

Evaluation of effects of PSMLE on oxidative stress and antioxidant markers

The serum malondialdehyde (MDA) levels of the rats were determined following the modified thiobarbituric acid method (Draper and Hadley, 1990). The serum was first mixed with trichloroacetic acid (TCA) and centrifuged to obtain the protein-free supernatant, to which was added 1% thiobarbituric acid and incubated for one hour at 95°C. The free MDA present in the supernatant reacted with thiobarbituric acid (TBA) to generate an MDA-TBA adduct, which was quantified colorimetrically at 532 nm wavelength using a SpectrumLab[®] spectrophotometer (HME Global Medical, England).

The serum catalase (CAT) activity was determined by the visible light method (Weydert and Cullen, 2010), using ElabScience catalase assay kit (ElabScience Biotechnology Co., Ltd., South Africa). In this determination, the decomposition of H_2O_2 by catalase in the plasma sample was quickly stopped by ammonium molybate, and the rest of H_2O_2 reacted with the ammonium molybate to generate a yellowish complex. The absorbance of the yellowish complex was measured at the wavelength of 405 nm using a Chem5V3[®] Semi-automated Clinical Chemistry Analyzer, (Erba Diagnostics, Mannheim GmbH, Mannheim, Germany) and the catalase activity was calculated from the absorbance.

The superoxide dismutase (SOD) activity was determined on serum, using the ElabScience SOD Assay kit (ElabScience Biotechnology Company Ltd., South Africa), which was based on the hydroxylamine method that adopted xanthine oxidase to measure SOD activity (Weydert and Cullen, 2010) with a Chem5V3[®] Semi-automated Clinical Chemistry Analyzer, (Erba Diagnostics, Mannheim GmbH, Mannheim, Germany), set at 550 nm wavelength.

The glutathione peroxidase (GPx) activity was determined on heparinized whole blood using a Fortress Diagnostics GPx test kit (Fortress Diagnostics Ltd, Antrim, UK.), based on the method described by Weydert and Cullen (2010). In this determination, the GPx in blood catalysed the oxidation of glutathione (GSH) by cumene hydroperoxide. The oxidized glutathione was converted to the reduced form in the presence of glutathione reductase and NADPH, and the NADPH was oxidized to NADP⁺ simultaneously. The absorbance of the solution at 340 nm wavelength was measured using a Chem5V3[®] Semi-automated Clinical Chemistry Analyzer, (Erba Diagnostics, Mannheim GmbH, Mannheim Germany) and the GPx activity concentration was calculated from the absorbance.

The serum total glutathione (GSH) was determined using RayBio[®] Glutathione Colorimetric Detection Kit (RayBiotech Inc., Georgia, USA), which was based on the modified enzyme recycling system by 5, 5-dithiobis–(2-nitrobenzoic acid) (DNTB) and glutathione reductase (Tipple and Rogers, 2012). The DNTB and glutathione reacted to generate 2-nitro-5-thiobenzoic acid which has a yellow color. The glutathione concentration was determined by measuring the absorbance of the generated yellow solution at 412 nm using Diatek[®] DR-3508G Microplate (ELISA) Reader, (Wuxi Hiwell Diatek Instruments Co. Ltd., China). The total glutathione concentration in the plasma was obtained from the standard total glutathione calibration curve.

Measurement of the liver weight and the calculation of the relative liver weight (liver weight percentage of body weight)

After blood sample collection, the rats were euthanized by intraperitoneal injection of 250 mg/kg Thiopentone sodium and confirmatory exsanguination (AVMA, 2013). The liver of each rat was carefully eviscerated and weighed, and the relative liver weight (liver weight percentage of the body weight) was calculated.

Evaluation of the *in vitro* antioxidant activity (AA) of PSMLE using (DPPH) assay method

The *in vitro* anti-oxidant activity of PSMLE was analyzed following the 1, 1-diphenyl 2-picryl hydrazyl (DPPH) assay method (Mensor et al., 2001). Two milliliters of PSMLE in distilled water at concentrations of 10, 50, 100, 200 and 400 µg/ml were each mixed with 1 ml of 0.5 mM DPPH (in methanol) in a cuvette. The DPPH reagent formed yellow coloured complexes with the free hydroxyl group present in the crude extract. The absorbance of the coloured complex was read at 517 nm wavelength after 30 minutes of incubation in the dark at room temperature, with the aid of a spectrophotometer (HME Global Medical, England). The tests were done in triplicates. A mixture of 1 ml of methanol and 2 ml of PSMLE served as blank, while 1 ml of 0.5 mM DPPH solution and 2 ml of methanol served as negative control. Ascorbic acid was used as the reference standard. The percentage AA was calculated as follows:

% AA= 100 - [(Absorbance of sample - Absorbance of blank) × 100] /Absorbance of control (Mensor et al., 2001; Oke and Hamburger, 2002).

Data analysis

Data obtained from the *in vivo* experiment with rats, and the results of PSMLE and ascorbic acid across the different concentrations, were subjected to one way analysis of variance (ANOVA). Variant means were separated post-hoc using the least significant difference method. Significance was accepted at p < 0.05. The *in vitro* anti-oxidant activities of PSMLE at the varied concentrations were compared with that of ascorbic acid using the student's t-test. A summary of the results are presented as bar charts with standard deviation bars and as table of means with standard error.

Rat groups		% change in body			
	Day 0	Day 3	Day 7	Day 14	weights
Group 1	183.17 ± 8.02	186.37 ± 8.09	187.57 ± 8.51	189.47 ± 8.45	3.42 ± 1.31
Group 2	181.23 ± 10.96	182.57 ± 10.63	184.20 ± 10.34	187.73 ± 8.92	3.83 ± 1.30
Group 3	179.73 ± 6.06	182.10 ± 6.30	182.33 ± 4.91	186.40 ± 4.73	3.71 ± 0.83
Group 4	182.36 ± 7.84	184.03 ± 6.93	184.88 ± 8.47	189.41 ± 7.52	3.85 ± 1.29
Group 5	180.30 ± 8.19	181.43 ± 7.74	183.78 ± 8.01	187.58 ± 7.57	4.04 ± 1.40

Table 1. Mean ± standard error of the body weights (g) and percentage change in body weights of albino rat groups* given graded acute oral doses of *Pterocarpus santalinoides* methanol leaf extract (PSMLE).

No significant differences (p > 0.05) in the body weight and percentage change in body weight of the rats groups. * Groups: Group 1 - 0 mg/kg PSMLE (untreated); Group 2 - 50 mg/kg PSMLE; Group 3 - 300 mg/kg PSMLE; Group 4 - 2000 mg/kg PSMLE; and Group 5 - 5000 mg/kg PSMLE.

Table 2. Mean ± standard error of the organ weights (g) of albino rat groups* given graded acute oral doses of *Pterocarpus santalinoides* methanol leaf extract (PSMLE).

Rat groups	Organ weights (g)					
	Liver weight	Average kidney weight	Spleen weight	Heart weight		
Group 1	6.85 ± 0.34	0.6l ± 0.02	0.72 ± 0.05	0.63 ± 0.05		
Group 2	6.76 ± 0.22	0.60 ± 0.05	0.70 ± 0.03	0.62 ± 0.04		
Group 3	6.69 ± 0.23	0.58 ± 0.01	0.69 ± 0.05	0.65 ± 0.03		
Group 4	6.78 ± 0.24	0.60 ± 0.06	0.71 ± 0.04	0.63 ± 0.04		
Group 5	6.74 ± 0.21	0.59 ± 0.02	0.67 ± 0.05	0.62 ± 0.05		

No significant differences (p > 0.05) in the organ weights of the rat groups. * Groups: Group 1 – 0 mg/kg PSMLE (untreated); Group 2 – 50 mg/kg PSMLE; Group 3 – 300 mg/kg PSMLE; Group 4 – 2000 mg/kg PSMLE; and Group 5 - 5000 mg/kg PSMLE.

RESULTS

Acute toxicity

There was no sign of toxicity or abnormality and no mortality in all the rat groups given the varied acute doses (0, 50, 300, 200 and 5000 mg/kg) of PSMLE all through the 14 day observation/monitoring period. All the rats which were given the varied acute doses were normal, with no sign of changes in the behavioral pattern, mucous membrane, skin and eyes. There were no significant differences (p > 0.05) in the body weights and the percentage change in body weights of the rat groups treated with the varied acute doses of PSMLE when compared to the untreated control (Table 1). There were also no significant differences (p > 0.05) between the groups given the acute doses of PSMLE and the untreated in their organ weights and relative organ weights (organ weight percentages of body weights) (Tables 2 and 3).

Phytochemical analysis

Phytochemical analysis revealed the presence of high

(+++) levels of tannins, glycosides, carbohydrates, saponins, phenols, sterols and terpenes; moderate (++) levels of flavonoids and low (+) level of alkaloid (Table 4).

Effects of PSMLE on blood levels of markers of hepatotoxicity

Treatment with 250 and 500 mg/kg PSMLE (Groups C and D) led to significantly (p < 0.05) lower serum ALT activity (Table 5). The 250 mg/kg PSMLE dose also led to significantly (p < 0.05) lower serum AST activity when compared to the negative control (Group A), and compared favorably with that of Group E (100 mg/kg silymarin) and Group F (normal control) rats (Table 5). There was no significant (p > 0.05) differences in the ALP activity between all the PSMLE-treated groups (Groups B, C and D) and the negative control (Group A), though the recorded lowering of the ALP activity in the treated groups occurred in a dose dependent manner (Table 5). The ALP activity of the normal control (Group F) was significantly (p < 0.05) lower than that of the negative control and all the PSMLE-treated groups (Table 5). Treatment with PSMLE at 250 mg/kg (Group C) led to

Det erreune	Relative organ weights (%)					
Rat groups	Liver weight %	Kidney weight %	Spleen weight %	Heart weight %		
Group 1	3.54 ± 0.03	0.32 ± 0.01	0.37 ± 0.02	0.33 ± 0.01		
Group 2	3.61 ± 0.05	0.32 ± 0.01	0.38 ± 0.02	0.32 ± 0.01		
Group 3	3.59 ± 0.05	0.31 ± 0.02	0.37 ± 0.02	0.35 ± 0.02		
Group 4	3.60 ± 0.04	0.32 ± 0.01	0.38 ± 0.02	0.33 ± 0.02		
Group 5	3.60 ± 0.04	0.32 ± 0.01	0.36 ± 0.02	0.33 ± 0.02		

Table 3. Mean ± standard error of the organ weight percentages (relative organ weights) of albino rats groups* given graded acute oral doses of *P. santalinoides* methanol leaf extract (PSMLE).

No significant differences (p > 0.05) in the relative organ weights of the rats groups. * Groups: Group 1 - 0 mg/kg PSMLE (untreated); Group 2 - 50 mg/kg PSMLE; Group 3 - 300 mg/kg PSMLE; Group 4 - 2000 mg/kg PSMLE; and Group 5 - 5000 mg/kg PSMLE.

 Table 4. Phytochemical constituents of P. santalinoides methanol leaf extract (PSMLE).

Phytochemical constituents	Semi-quantitative composition		
Tannins	+++		
Terpenes and sterols	+++		
Alkaloids	+		
Flavonoids	++		
Saponins	+++		
Carbohydrates	+++		
Glycosides	+++		
Phenol	+++		

+++ = High levels present; ++ = Moderate levels present; + = Low levels present.

Table 5. Effects of oral administration of graded doses of PSMLE on serum enzyme activities of albino rats given sub-acute toxic doses of CCI_4

C	Ν	lean ± standard error (U/	L)
Group	ALT	AST	ALP
Group A	102.4 ± 4.79 ^a	121.03 ± 8.03^{a}	384.46 ± 60.81 ^a
Group B	98.78 ± 6.83 ^a	120.82 ± 7.41 ^a	391.01 ± 46.11 ^a
Group C	48.36 ± 7.02 ^b	77.28 ± 7.96 ^b	358.42 ± 10.08 ^a
Group D	79.11 ± 6.08 [°]	106.92 ± 5.16 ^a	338.41 ± 18.93 ^a
Group E	46.96 ± 7.48 ^b	75.03 ± 6.74 ^b	279.97 ± 22.86 ^{ab}
Group F	30.83 ± 4.61 ^b	63.01 ± 6.42 ^b	104.11 ± 21.04 ^b

^{a, b, c} Different alphabetical superscripts in a column indicate significant (p < 0.05) differences between the groups.

*Groups: Group A – CCl₄ + 10 ml/kg distilled water (negative control); Group B – CCl₄ + 50 mg/kg PSMLE; Group C – CCl₄ + 250 mg/kg PSMLE; Group D – CCl₄ + 500 mg/kg PSMLE; Group E – CCl₄ + 100 mg/kg silymarin; Group F – No CCl₄ and no treatment.

significantly (p < 0.05) higher serum total protein levels when compared to the negative control (Group A) (Figure 2). Treatment with PSMLE at all the doses used in the study had no significant (p > 0.05) effect on the serum albumin levels (Figure 2). There was significantly (p < 0.05) 0.05) higher serum globulin level in albino rats treated with 250 mg/kg PSMLE (Group C) when compared to the negative control (Group A). There was no significant (p > 0.05) differences in serum globulin levels between 250 mg/kg PSMLE-treated rats (Group C), 100 mg/kg



Figure 2. Effects of oral administration of graded doses of PSMLE on serum proteins of albino rats given subacute toxic doses of CCl₄; [Group A – CCl₄ + 10 ml/kg distilled water (negative control); Group B - CCl₄ + 50 mg/kg PSMLE; Group C - CCl₄ + 250 mg/kg PSMLE; Group D - CCl₄ + 500 mg/kg PSMLE; Group E - CCl₄ + 100 mg/kg Silymarin; Group F – No CCl₄, No treatment (normal control)].

silymarin-treated (Group E) and the normal control (Group F) (Figure 2).

The serum total cholesterol level of Group D rats (500 mg/kg PSMLE) was significantly (p < 0.05) lower than that of the negative control (Group A), and there were no significant (p > 0.05) differences in serum total cholesterol levels between Group D and Groups B, C, E and F (Table 6). Treatment with 250 mg/kg PSMLE (Group C) and 100 mg/kg silymarin (Group E) significantly (p < 0.05) lowered serum total bilirubin levels when compared to the negative control (Group A) and other PSMLE-treated groups (Table 6). There was no significant (p > 0.05) difference in serum total bilirubin level between the rat groups treated with 250 mg/kg PSMLE, 100 mg/kg silymarin and the normal control (Table 6). The liver weight of the rat groups treated with 50, 250 and 500 mg/kg PSMLE and 100 mg/kg silvmarin (Groups B, C, D and E) were significantly (p < 0.05) lower than that of the negative control (Group A), with Groups C, D and E being significantly (p < 0.05) lower than Group B, and comparing favorably with the normal control (Group F) (Figure 3). Treatment with PSMLE at all doses used in the study (Groups B, C and D) and 100 mg/kg silymarin (Group E) led to significantly (p < 0.05) lower relative liver weights of the albino rats when compared to the negative control (Group A) (Figure 3).

Effects of PSMLE on oxidative stress and antioxidant markers

Treatment with PSMLE caused a dose-dependent reduction in MDA levels across the groups, with the negative control (Group A) being significantly (p < 0.05) higher than 250 and 500 mg/kg PSMLE, 100 mg/kg silymarin and normal control (Groups C, D, E and F), but not significantly (p > 0.05) different from 50 mg/kg PSMLE (Group B) (Table 7). The CAT activity of Groups B, C, D and E was significantly (p < 0.05) higher than that of Group A rats, and there were no significant (p > 0.05)

0	Mean ± standard error (mg/dl)			
Group	Total cholesterol	Total bilirubin		
Group A	81.23 ± 8.04 ^a	0.70 ± 0.06 ^a		
Group B	63.48 ± 10.84 ^{ab}	0.63 ± 0.14 ^a		
Group C	61.64 ± 3.02 ^{ab}	0.36 ± 0.07^{b}		
Group D	56.48 ± 2.42 ^b	0.66 ± 0.03^{a}		
Group E	64.19 ± 5.76^{ab}	0.37 ± 0.04^{b}		
Group F	62.49 ± 3.54^{ab}	0.38 ± 0.03^{b}		

Table 6. Effects of oral administration of graded doses of PSMLE on serum total cholesterol and serum total bilirubin levels of albino rats given sub-acute toxic doses of CCl₄.

^{a, b, c} Different alphabetical superscripts in a column indicate significant (p < 0.05) differences between the groups. *Groups: Group A – CCl₄ + 10 ml/kg distilled water (negative control); Group B – CCl₄ + 50 mg/kg PSMLE; Group C – CCl₄ + 250 mg/kg PSMLE; Group D – CCl₄ + 500 mg/kg PSMLE; Group E – CCl₄ + 100 mg/kg silymarin; Group F – No CCl₄ and no treatment.



Figure 3. Effects of oral administration of graded doses of PSMLE on liver weight and relative liver weight of albino rats given sub-acute toxic doses of CCl₄. [Group A – CCl₄ + 10 ml/kg distilled water (negative control), B - CCl₄ + 50 mg/kg PSMLE; Group C - CCl₄ + 250 mg/kg PSMLE; Group D - CCl₄ + 500 mg/kg PSMLE; Group E - CCl₄ + 100 mg/kg Silymarin; Group F – No CCl₄. No treatment (normal control)]

differences between Groups B, C, D and E in their CAT activity (Table 7). The SOD activity was also significantly (p < 0.05) lower in Groups A and B compared to other

groups (Table 7). Glutathione peroxidase (GPx) activity followed the same pattern as SOD with Groups A and B being significantly (p < 0.05) lower than Groups C, D, E

Table 7. Effects of oral administration of graded doses of PSMLE o	n in vivo oxidative stress and blood anti-oxidant marker levels of albino
rats given sub-acute toxic doses of CCl _{4.}	

Oxidative stress and	Means, with standard deviation in brackets						
antioxidant parameter	Group A	Group B	Group C	Group D	Group E	Group F	
MDA (µmol/L)	17.61 (3.32) ^a	14.19(4.29) ^{ab}	9.65 (2.87) ^{bc}	7.52 (3.30) ^{cd}	7.65(2.88) ^{cd}	4.05 (1.90) ^d	
CAT (IU/ml)	12.18(2.36) ^a	21.55(2.38) ^b	20.92 (3.35) ^b	20.35(3.49) ^b	23.14(4.40) ^{bc}	29.76(7.99) ^c	
SOD (IU/ml)	0.221 (0.020) ^a	0.257(0.049) ^{ab}	0.300(0.024) ^c	0.278(0.020) ^{bc}	0.294 (0.027) ^{bc}	0.299 (0.019) ^c	
GPx (IU/L)	82.50 (10.74) ^a	88.41(12.89) ^a	119.35 (20.17) ^b	124.29(23.90) ^b	127.16 (17.27) ^b	190.25(18.58) ^c	
GSH (µg/µl)	0.016 (0.006) ^a	0.029(0.004) ^b	0.039(0.005) ^{bc}	0.041(0.005) ^{bc}	0.046 (0.006) ^c	0.068(0.015) ^d	

^{a,b,c} Different alphabetical superscripts in a row indicate significant differences between the groups (p < 0.05). *Groups: Group A – CCl₄ + 10 ml/kg distilled water (negative control); Group B – CCl₄ + 50 mg/kg PSMLE; Group C – CCl₄ + 250 mg/kg PSMLE; Group D – CCl₄ + 500 mg/kg PSMLE; Group E – CCl₄ + 100 mg/kg silymarin; Group F – No CCl₄ and no treatment.

and F (Table 7). Treatment with PSMLE at all doses (50, 250, 500 mg/kg) and silymarin (100 mg/kg) (Groups B, C, D and E) led to significantly (p < 0.05) higher GSH levels when compared to the negative control (Group A) (Table 7). There was significant (p < 0.05) dose-dependent higher levels of GSH when all the PSMLE-treated groups were compared. However, there were significantly (p < 0.05) lower GSH levels in all the PSMLE-treated groups when compared with the normal control (Group F) (Table 7).

In vitro antioxidant activity of PSMLE

The antioxidant activity of PSMLE at 10 µg/ml concentration was significantly (p < 0.05) lower than that of ascorbic acid (Figure 4). However, 50, 100, 200 and 400 µg/ml concentration of PSMLE produced significantly (p < 0.05) higher antioxidant activity when compared to the same concentrations of ascorbic acid (Figure 4). The PSMLE showed maximum anti-oxidant activity at 400 µg/ml and minimum activity at 10 µg/ml. There were no significant (p > 0.05) differences between the anti-oxidant activity values obtained at 50 µg/ml, 100 µg/ml, and 200 µg/ml concentrations (Figure 4). Also, there were no significant (p > 0.05) differences between the anti-oxidant activity values of the ascorbic acid standard at varied concentrations (Figure 4).

DISCUSSION

In the acute toxicity study, the rats tolerated the extract up to 5000 mg/kg, without any significant/adverse changes in their body and organ weights. This shows that the LD_{50} is above 5000 mg/kg (OECD, 2001). An LD_{50} above 5000 mg/kg is within the World Health Organization's category of substances "unlikely to present acute hazard in normal use" (WHO, 2001). This implies that PSMLE is safe for acute use in the treatment of ailments and diseases for which it is effective (OECD, 2001). This concurs with earlier reports by Ihedioha et al. (2017; 2018) that methanol leaf extracts of *P. santalinoides* is not acutely toxic.

The phytochemicals observed in PSMLE are essential bioactive compounds commonly found in various herbs used for medicinal purposes. Similar compounds as recorded in this study have been identified in the methanol and ethanol extracts of the leaf and stem bark of *P. santalinoides* (Anowi et al., 2012; Eze et al., 2012; Odeh and Tor-Anyim, 2014; Enemali et al., 2019).

The higher serum ALT, AST and ALP activities in all the groups that were given CCl₄ showed that CCl₄ administration damaged the liver cells and altered the integrity of the hepatocytes (Boll et al., 2001; Kim et al., 2010). Elevation in serum transaminases (AST and ALT) a biomarker of hepatocellular necrosis is and hepatotoxicity (Friedman et al., 1996). The administration of PSMLE at the doses of 250 and 500 mg/kg led to significantly lower serum ALT activity, and at 250 mg/kg it led to significantly lower serum AST activity, and these were comparable to that of silymarin treatment (a known hepatoprotective drug). These findings are in agreement with the reports of Offor et al. (2015) who also recorded significantly lower ALT and AST activities in CCI₄-induced hepatotoxic albino rats treated with ethanol leaf extract of *P. santalinoides*. Reports by Ihedioha et al. (2017) also showed that methanol leaf extract of P. santalinoides restored hepatocellular integrity in acetaminopheninduced hepatotoxicity in albino rats. In another study on a related species of Pterocarpus, Ihedioha et al. (2019) also reported significantly lower serum ALT and AST activities in CCl₄-induced hepatotoxic albino rats treated with methanol leaf extract of Pterocarpus mildbraedii (a plant species in the same Genus Pterocarpus). The ability of the PSMLE to protect hepatocellular integrity from CCl₄-induced damage is believed to be due to its phytochemical constituents such as tannins, flavonoids and glycosides (Muriel et al., 2001; Ihedioha et al., 2017), which have been reported to be hepatoprotective. Other



Figure 4. The *in vitro* antioxidant activity values of varied concentrations of PSMLE and ascorbic acid using 1, 1-diphenyl 2 picryl hydrazyl (DPPH) assay.

natural products that possess similar phytochemical constituents have also been reported to protect against CCl₄-induced liver damage (Hsiao et al., 2003).

The significantly lower serum total proteins and serum globulin in the negative control group and other groups that were given CCl_4 are indications of the ability of CCl_4 to damage hepatocytes and impair protein synthesis (Navarro and Senior, 2006). Proteins form the major portion of dissolved substances in the plasma (Singh et al., 2011), and act as transport agents for a wide variety of substances such as hormones, lipids and vitamins. Proteins can be used as a supplementary test for hepatic biosynthetic functions (Friedman et al., 1996; Thapa and Walia, 2007; Singh et al., 2011). Decreased levels of proteins are found in liver disorders (Kipple, 2003). The

PSMLE (250 mg/kg) ameliorated the impaired protein synthetic function by leading to significantly higher serum total protein and serum globulins of the treated rats.

Elevation of serum globulin in CCl_4 -induced hepatotoxic albino rats treated with *P. mildbraedii* (a plant of the same genus as *P. santalinoides*) methanol extract has also been reported (Ihedioha et al., 2019). Stimulation of protein synthesis is known to be a hepatoprotective mechanism. Protein synthesis accelerates the hepatocyte regeneration process and helps in the production of replacement liver cells (Rip et al., 1985; Tadeusz et al, 2001).

The significantly higher serum total cholesterol in the negative control group and other groups that were given CCl_4 are indications of alterations in serum lipid profile

caused by CCl₄ administration. The liver is the major site for the synthesis and clearance of lipoproteins; therefore, hepatotoxicity or damage to the hepatocytes can affect plasma lipids leading to alterations in lipid profile, and this has been found to be instrumental to the development of atherosclerosis (Ihedioha et al., 2013; Ihedioha et al., 2018). Hypercholesterolaemia occurs in hepatotoxicity as consequent upon dysfunction dyslipidaemia and alterations in hepatic lipid synthesis and clearance (Mandal et al., 2013). Treatment with PSMLE (500 mg/kg) and silymarin (100 mg/kg) ameliorated these effects by significantly lowering the serum total cholesterol of the treated albino rats at these doses. Ihedioha et al. (2017) also reported significantly lower serum total cholesterol levels in acetaminophen-induced hepatotoxic albino rats treated with methanol leaf extract of P. santalinoides.

Administration of CCl₄ also adversely affected bilirubin excretion as observed in all the groups that were given CCl₄ Bilirubin is an endogenous anion derived from the regular degradation of haemoglobin from the red blood cells. It is a chemical normally present in the blood in small amounts and excreted from the liver in form of bile. When the liver cells are damaged, they may not be able to excrete bilirubin in the normal way, thus causing a build-up of bilirubin in the blood and extracellular fluid (Singh et al., 2011). Increased levels of bilirubin may also result from decreased hepatic clearance and lead to jaundice and other hepatotoxicity symptoms (Saukkonen et al., 2006). Treatment with PSMLE (500 mg/kg) and silymarin (100 mg/kg) led to significantly lower serum total bilirubin in albino rats given toxic doses of CCl₄. They ameliorated the impaired excretory function induced by the CCl₄ administration, and enhanced hepatic clearance of bilirubin. Ihedioha et al. (2017) reported that P. santalinoides methanol extract significantly lowered bilirubin level acetaminophen-induced total in hepatotoxicity in albino rats. The ethanol extract of a related species (Pterocarpus milbraedii) had been reported by other researchers to significantly lower serum total bilirubin level in propanil and CCl4-induced hepatotoxicity in albino rats (Otuechere and Farombi, 2015; Hamza et al., 2017; Ihedioha et al., 2019).

The relatively higher liver weights in albino rats given CCl_4 are indications of inflammation and/or degeneration which accompany CCl_4 toxicity (Tahashi et al., 2002; Bukhsh et al., 2014). Inflammation is a complex catalogue of vascular and tissue changes that develop as a response of tissue to injury (Ihedioha, 2003). Degeneration on the other hand, is a retrogressive change in tissues characterized by abnormal changes and decreases in function. Organs undergoing acute degenerative changes tend to be larger and heavier than normal (Ihedioha, 2003). Enlargement of the liver (hepatomegaly) is a common evidence of hepatic injury (Ihedioha and Chineme, 2005). Treatment with PSMLE at

all the doses used in the study and silymarin (100 mg/kg) significantly lowered this inflammation/degenerationinduced enlargement caused by CCl₄ toxicity and this suggests their amelioration of this inflammatory enlargement of the liver. Relatively lower liver weight has also been reported in hepatotoxic albino rats treated with *P. mildbraedii* methanol leaf extract (Ihedioha et al., 2019).

In the in vivo oxidative stress and antioxidants evaluation results, the higher doses of PSMLE (250 mg/kg and 500 mg/kg) were able to mop up more free radicals more than the lower dose (50 mg/kg). This implies that the PSMLE possesses anti-oxidant activity which is dose-related. Malondialdehvde (MDA) is one of the final products of the peroxidation of polyunsaturated fatty acids in cells and is a known marker of oxidative stress status in several diseases (Draper and Hadley, 1990; Gawel et al., 2004; DelRio et al., 2005). The MDA result in this present study concurs with the reports of Maruthupandian and Mohan (2011) on the lowering of plasma lipid peroxides in diabetic rats given ethanol extract of Pterocarpus marsupium wood and bark (a plant belonging to the related same aenus). Administration of PSMLE enhanced catalase (CAT) activity. This is evident by the significantly higher catalase activity in all the groups treated with PSMLE. Catalase is a very important enzyme in protecting the cell from oxidative damage (Chelikani et al., 2004). It is the main enzyme that removes hydrogen peroxide (a reactive oxygen species), and blocks oxidative stress (Gaetani et al., 1996). The findings in this present study are similar to the reports by Maruthupandian and Mohan (2011) on the effects of a related plant of the same genus (Pterocarpus marsupium) on catalase activity of alloxan-induced diabetic rats. The superoxide dismutase (SOD) assay result in this present study is also an indication that the negative control (Groups A) and 50 mg/kg PSMLEtreated (Group B) rats had no significant protection from oxidative damage as compared to other groups. Treatment with PSMLE at the doses of 250 and 500 mg/kg (Groups C and D) effectively restored the SOD activity of these rats to a level comparable to that of normal rats not given CCl₄ (Group F). Superoxide dismutase (SOD) is an important anti-oxidant defense in living cells exposed to oxygen radicals. Earlier studies had shown that treatment with SOD decreased the generation of reactive oxygen species and oxidative stress (Fridovich, 1997; Gongora et al., 2006). Treatment with PSMLE at the doses of 250 and 500 mg/kg also restored the glutathione peroxidase (GPx) activity of the albino rats. The main biological role of GPx is to protect the organism from oxidative damage. Its biochemical function is to reduce lipid hydroperoxides to their corresponding alcohols and to reduce free hydrogen peroxide to water. Blood levels of GPx have been reported to be low in some important diseases, and had

significantly improved when antioxidants were administered (Lubos et al., 2011). The findings in this present study of significantly higher GPx activity in PSMLE-treated groups is comparable to the reports by Maruthupandian and Mohan (2011) on the effects of P. marsupium wood and bark extracts on GPx of alloxaninduced diabetic rats. Plasma total glutathione (GSH) is one of the major endogenous anti-oxidants produced by cells, which participates directly in the neutralization of free radicals and reactive oxygen species, as well as maintaining exogenous anti-oxidants such as vitamins C and E in their reduced (active) forms (Dringen, 2000). The higher total glutathione level recorded in this study for Group B (50 mg/kg PSMLE), Group C (250 mg/kg PSMLE), and Group D (500 mg/kg PSMLE) implies a dose-dependent anti-oxidant activity in these groups. Maruthupandian and Mohan (2011) also reported the enhancing effects by P. masurpium extracts on the plasma total glutathione of alloxan-induced diabetic rats, but their results did not show dose-dependence.

In vitro, the PSMLE evoked significant concentrationdependent antioxidant activity, from 50 to 400 μ g/ml concentrations, suggesting that it could be of benefit in ameliorating tissue damaging effect of increased concentration of reactive oxygen species (ROS) seen in toxic liver diseases. This result concurs with the reports of Bothon et al. (2014), Kabine et al. (2015) and Akaniro-Ejim et al. (2018) who also reported that the aqueous and hydro-ethanol leaf and fruit extracts, and aqueousethanol leaf extract of *P. santalinoides* respectively, possess anti-oxidant activity *in vitro*.

The antioxidant properties of PSMLE may be attributed to its flavonoids, tannins and/or glycoside contents (Pietta, 2000; Hodek et al., 2002; Yi-Fang et al., 2002; Aruoma, 2003). The results obtained from the present study also concurred with the reports by lhedioha et al. (2017), who attributed the ability of PSMLE in lowering serum ALT and AST activities to its antioxidant activity. Ihedioha et al. (2018) also reported that the lipid lowering capability of aqueous leaf infusion of P. santalinoides may be related to its antioxidant properties. This present result is also in agreement with reports by other researchers on the in vitro antioxidant activity of other species in the same genus Pterocarpus, such as Pterocarpus marsupium (Mohammadi et al., 2009; Tippani et al., 2010; Maruthupadian and Mohan, 2011), Pterocarpus mildbraedii (Nwozo et al.. 2015). Pterocarpus angolensis (Traore et al., 2016) and Pterocarpus erinaceus (Patrick et al., 2016). These findings are significant because foods, spices and herbal formulations containing antioxidants are used pharmacologically to prevent, manage and /or treat diseases in which oxidative stress play critical roles such as liver diseases, heart diseases and diabetes mellitus (Sies, 1997; Halliwell and Gutteridge, 1989; Valko et al., 2007). The present study was limited to the evaluation of

the hepatoprotective and antioxidant activity of the crude methanol extract of the leaves of *P. santalinoides*. Further studies to elucidate the active fractions and pure compound(s) responsible for the hepatoprotective and antioxidant activity are ongoing, and shall be reported in future.

Conclusion

The findings in this present study confirm that extracts of the leaves of *P. santalinoides* significantly ameliorated hepatocellular injury and relieved oxidative stress, and may thus be beneficial in the treatment and management of toxic liver damage and diseases mediated by and/or associated with oxidative stress.

CONFLICT OF INTERESTS

The authors have not declared any conflict of interests.

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