

Full Length Research Paper

Gas chromatography mass spectrum and fourier-transform infrared spectroscopy analysis of methanolic extract of *Rosmarinus officinalis* leaves

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Received 20 April 2015; Accepted 3 June 2015

The aims of the study were to investigate the presence of phytochemical compounds from the leaves of *Rosmarinus officinalis*, using methanolic extraction and report the main functional components by using FT-IR. The phytochemical compound screened by GC-MS method. Seventeen bioactive phytochemical compounds were identified in the methanolic extract of *R. officinalis*. The identification of phytochemical compounds is based on the peak area, retention time molecular weight, molecular formula, chemical structure, MS Fragment ions and pharmacological actions. GC-MS analysis of *Rosmarinus officinalis* revealed the existence of the α -pinene, Camphene, Eucalyptol, 2-Methoxy-4-vinylphenol, 1-Oxaspiro[4,5]deca-3,6-diene,2,6,10,10-tetramethyl, 1-Oxaspiro[4,5] deca-3,6-diene,2,6,10,10-tetramethyl, Neocurdione, Isoaromadendrene epoxide, 1b,4a-Epoxy-2H-cyclopenta[3,4]cyclopropa[8,9] cycloundec., Cis-Vaccenic acid, Phenanthrenol,4b,5,6,7,8,8a,9,10-octahydro-4b,8,8-trimethyl-1, Galanthamine, Dibenz[a,c]cyclohexane,2,4,7-trimethoxy, 2,4a,7-Trihydroxy-1-methyl-8-methyleneqibb-3-ene. 1,10-carboxylic acid, Retinoic acid, 7,8,12-Tri-O-acetyl-3-desoxy-ingol-3-one and 4,6-Androstadien-3 β -ol-17-one,acetate. The FTIR analysis of *Rosmarinus officinalis* leaves proved the presence of Alkenes, Aliphatic fluoro compounds, Alcohols, Ethers, Carboxylic acids, Esters, Nitro compounds, Alkanes, Aldehydes, Ketones compounds. It contain chemical constitutions which may be useful for various herbal formulation as anti-inflammatory, analgesic, antipyretic, cardiac tonic and antiasthmatic.

Key words: FT-IR, GC-MS analysis, Leaves, Methanol, *Rosmarinus officinalis*.

INTRODUCTION

Rosmarinus officinalis, related to the lamiaceae family of plants. Rosemary is used as an antispasmodic in real colic and dysmenorrhoea, in relieving respiratory disorders and to stimulate growth of hair. Rosemary (*Rosmarinus officinalis*), which grows wild around Mediterranean area, is a species from Lamiaceae family.

It grows as a shrub or herbaceous plant with about 0.8 to 2m height (Atik bekkara et al, 2007). This plant likes dry and arid regions, hills and low mountains, calcareous, shale, clay and rocky substrates. The potent antioxidant properties of rosemary extracts have been attributed to its phenolic compounds, mainly rosmarinic acid and

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diterpenes carnosic acid and carnosol (Sergi and Leonor, 2002; Troncoso et al., 2005; Ameer et al., 2015). Rosemary extract relaxes smooth muscles and has choleric, hepatoprotective and antitumorigenic activity (Al-Sereiti et al., 1999). Recent research shows that rosemary extracts possess strong anticancer properties. In the last few years gas chromatography mass spectrometry has become firmly established as a key technological platform for metabolite profiling in plant (Fiehn et al., 2003; Sumner et al., 2003; Fernie et al., 2004; Kell et al., 2005; Robertson et al., 2005). GC-MS based metabolome analysis has profound applications in discovering the mode of action of drugs or herbicides and helps unravel the effect of altered gene expression on metabolism and organism performance in biotechnological applications. Infrared spectroscopy provides a useful method for herbal analysis and elucidate the compounds structures as well as for quantitative analysis of drugs Bunaciu et al., 2010; Cheng et al., 2010; Bunaciu et al., 2010; Huda et al., 2015). Fourier transform infrared spectrometry is a physico-chemical analytical technique and one of the most widely used methods to identify the structure of unknown composition or its chemical group, and the intensity of the absorption spectra associated with molecular composition or content of the chemical group (Surewicz et al., 1993; Imad et al., 2015). The present study involves an assessment using GC-MS and FT-IR spectroscopic techniques to investigate the authenticity of commercial sample of the herbal drug by analyzing their fingerprints.

MATERIALS AND METHODS

Collection and preparation of plant material

The leaves were purchased from local market in Hilla city, middle of Iraq. After thorough cleaning and removal of foreign materials, the leaves were stored in airtight container to avoid the effect of humidity and then stored at room temperature until further use.

Preparation of sample

About 20 g of the plant sample powdered were soaked in 100 ml methanol for 16 h in a rotatory shaker. Whatman No.1 filter paper was used to separate the extract of plant. The filtrates were used for further phytochemical analysis (Imad et al., 2015a). It was again filtered through sodium sulphate in order to remove the traces of moisture.

Gas chromatography mass spectrum analysis

The GC-MS analysis of the plant extract was made in a (QP 2010 Plus SHIMADZU) instrument under computer control at 70 eV (Imad et al., 2014; Imad et al., 2015b and Muhanned et al., 2015). About 1 µl of the methanol extract was injected into the GC-MS using a micro syringe and the scanning was done for 45 min. As the compounds were separated, they eluted from the column and entered a detector which was capable of creating an electronic

signal whenever a compound was detected. The greater the concentration in the sample, bigger was the signal obtained which was then processed by a computer. The time from when the injection was made (Initial time) to when elution occurred is referred to as the Retention time (RT). While the instrument was run, the computer generated a graph from the signal called Chromatogram. Each of the peaks in the chromatogram represented the signal created when a compound eluted from the Gas chromatography column into the detector. The x-axis showed the RT and the y-axis measured the intensity of the signal to quantify the component in the sample injected. As individual compounds eluted from the Gas chromatographic column, they entered the electron ionization (mass spectroscopy) detector, where they were bombarded with a stream of electrons causing them to break apart into fragments. The fragments obtained were actually charged ions with a certain mass. The M/Z (Mass/Charge) ratio obtained was calibrated from the graph obtained, which was called as the Mass spectrum graph which is the fingerprint of a molecule. Before analyzing the extract using gas chromatography and mass spectroscopy, the temperature of the oven, the flow rate of the gas used and the electron gun were programmed initially. The temperature of the oven was maintained at 100°C. Helium gas was used as a carrier as well as an eluent. The flow rate of helium was set to 1ml per minute. The electron gun of mass detector liberated electrons having energy of about 70eV. The column employed here for the separation of components was Elite 1(100% dimethyl poly siloxane). The identity of the components in the extracts was assigned by the comparison of their retention indices and mass spectra fragmentation patterns with those stored on the computer library and also with published literatures (Imad et al., 2015c; Mohammed et al., 2015).

Fourier transform infrared spectrophotometer (FTIR)

The powdered sample of the plant specimen was treated for FTIR spectroscopy (Shimadzu, IR Affinity 1, Japan). The sample was run at infrared region between 400 and 4000 nm (Ameer et al., 2015 and Huda et al., 2015).

RESULTS AND DISCUSSION

Gas Chromatography and Mass spectroscopy analysis of compounds was carried out in methanolic seed extract of *Rosmarinus officinalis*, shown in Table 1. The GC-MS chromatogram of the 17 peaks of the compounds detected was shown in Figure 1. Chromatogram GC-MS analysis of the methanol extract of *Rosmarinus officinalis* showed the presence of nine major peaks and the components corresponding to the peaks were determined as follows. The First set up peak were determined to be α-pinene Figure 2. The second peak indicated to be Camphene Figure 3. The next peaks considered to be Eucalyptol, 2-Methoxy-4-vinylphenol, 1-Oxaspiro[4,5]deca-3,6-diene,2,6,10,10-tetramethyl, 1-Oxaspiro[4,5]deca-3,6-diene,2,6,10,10-tetramethyl, Neocurdione, Isoaromadendrene epoxide, 1b,4a-Epoxy-2H-cyclopenta[3,4]cyclopropa[8,9] cycloundec., Cis-Vaccenic acid, Phenanthrenol,4b,5,6,7,8,8a,9,10-octahydro-4b,8,8-trimethyl-1, Galanthamine, Dibenz[a,c]cyclohexane,2,4,7-trimethoxy, 2,4a,7-Trihydroxy-1-methyl-8-methyleneqibb-3-ene. 1,10-carboxylic acid, Retinoic acid, 7,8,12-Tri-O-acetyl-3-

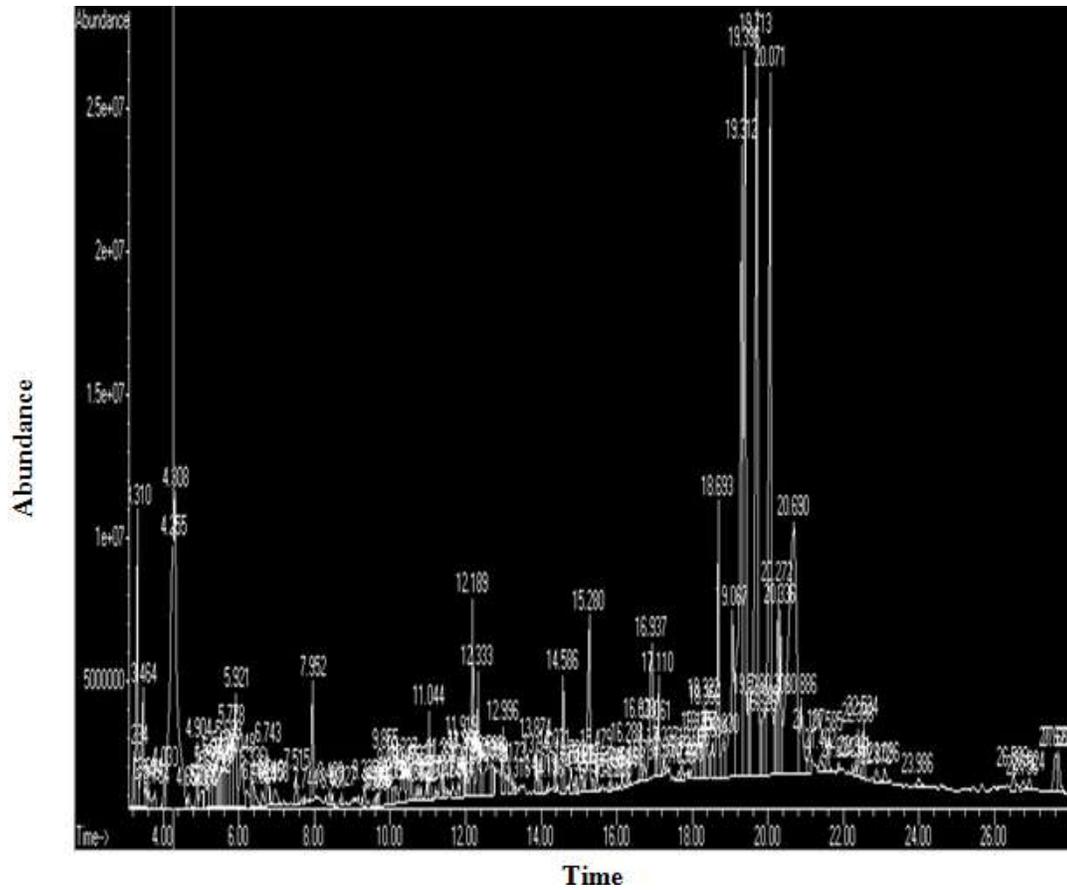


Figure 1. GC-MS profile of leaf extract of *Rosmarinus officinalis*.

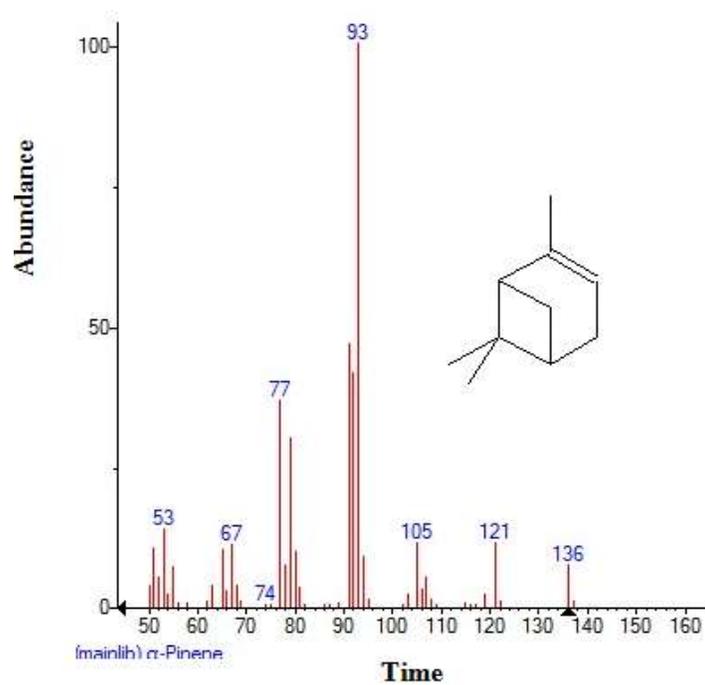


Figure 2. Structure of α -pinene present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

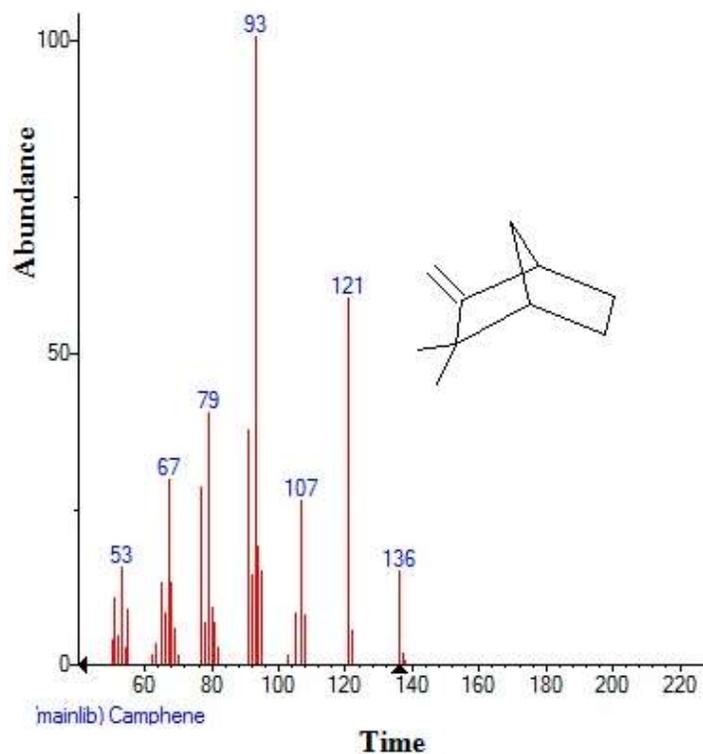


Figure 3. Structure of camphene present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

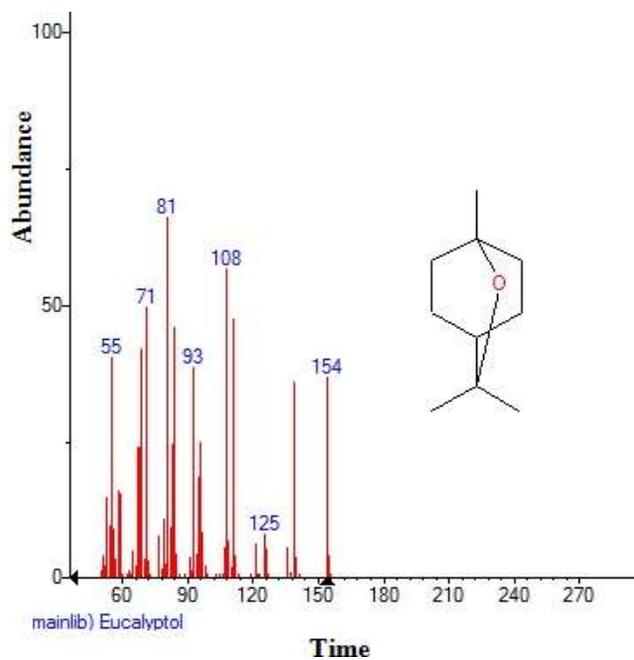


Figure 4. Structure of eucalyptol present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

desoxy-ingol-3-one, 4, 6- Androstadien-3 β -ol-17- one, acetate (Figures 4 to 17). FTIR analysis of dry

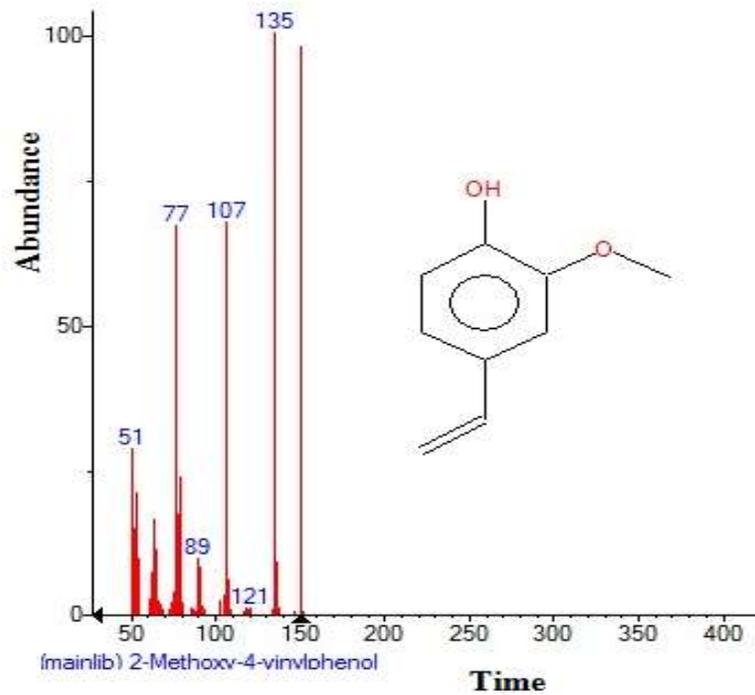


Figure 5. Structure of 2-Methoxy-4-vinylphenol present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

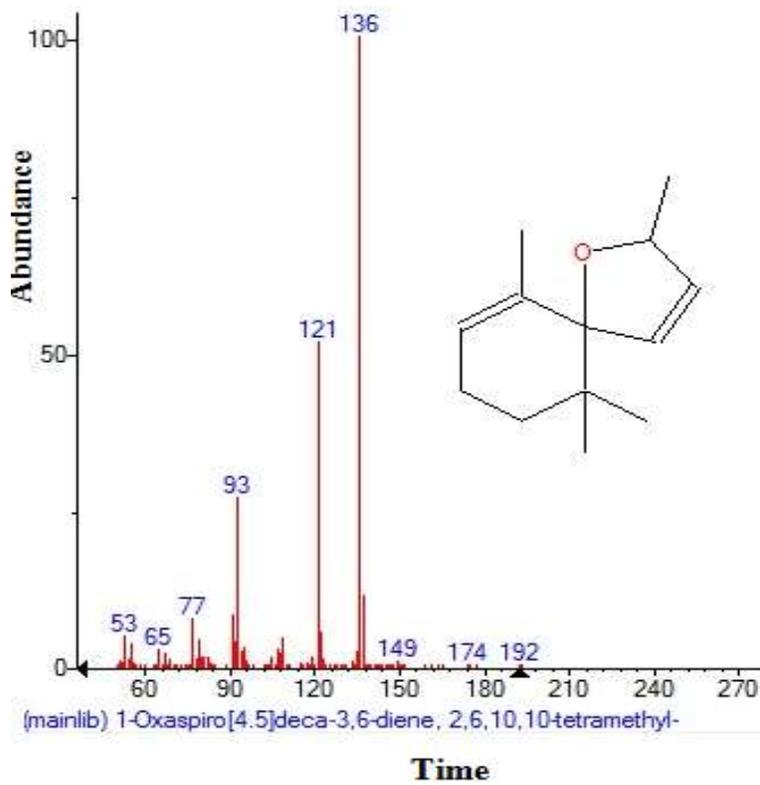


Figure 7. Structure of 1-Oxaspiro[4.5]deca-3,6-diene,2,6,10,10-tetramethyl present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

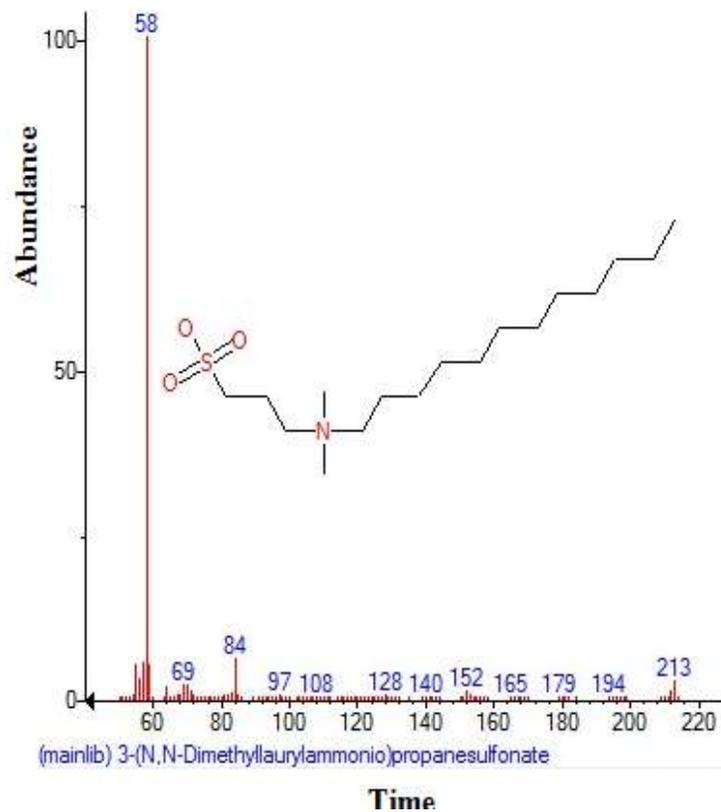


Figure 8. Structure of 3-(N,N-Dimethyl-laurylammonio)propanesulfate present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

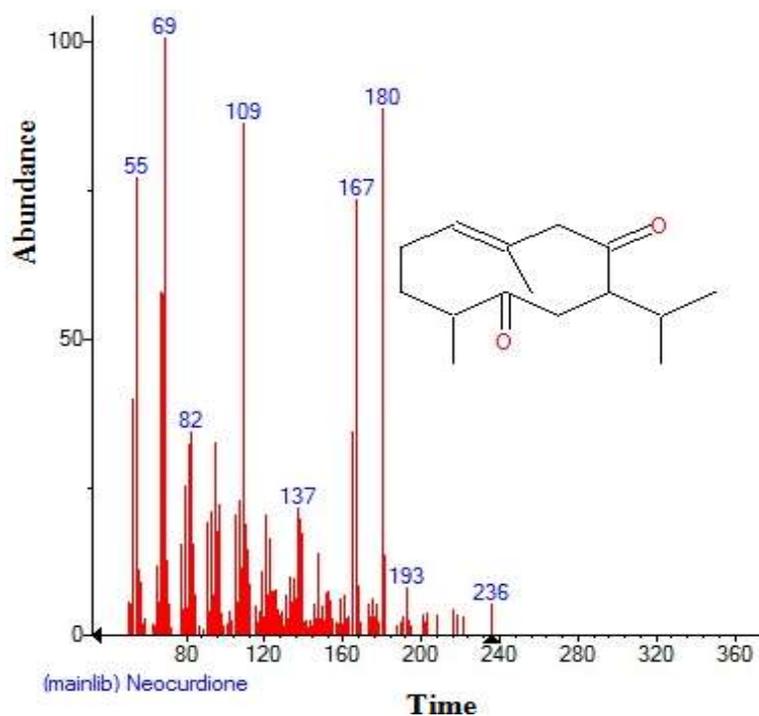


Figure 9. Structure of Neocurdione present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

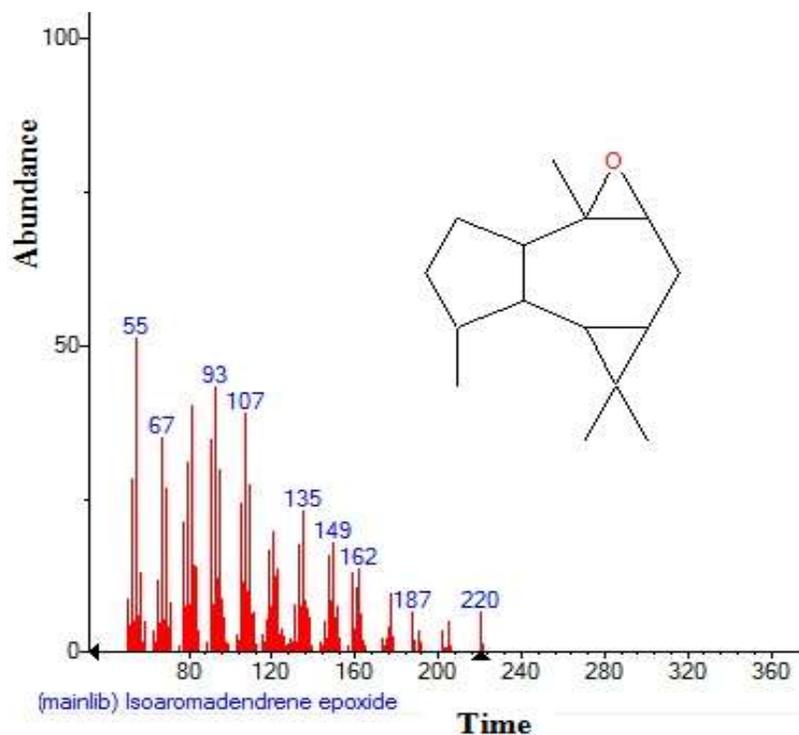


Figure 10. Structure of Isoaromadendrene epoxide present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

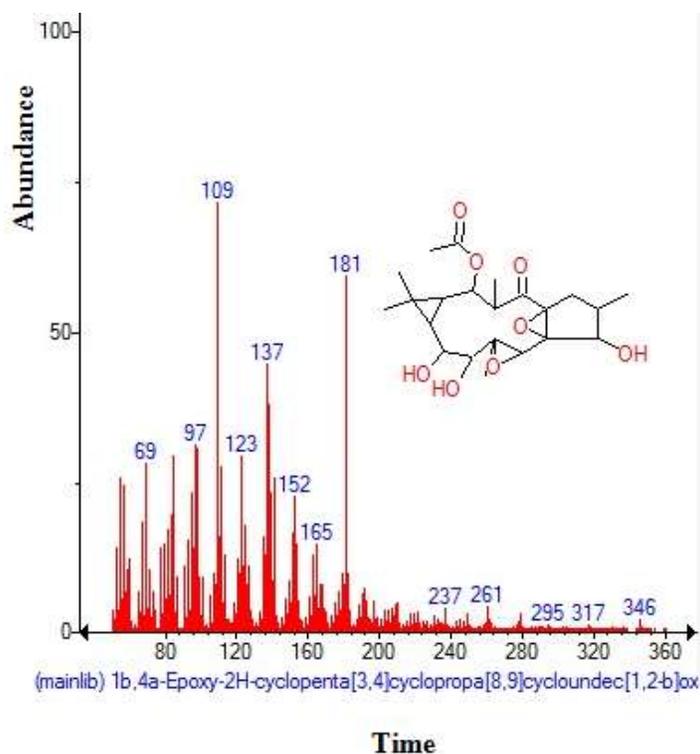


Figure 11. Structure of 1b,4a-Epoxy-2H-cyclopenta[3,4]cyclopropa[8,9]cycloundec. present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

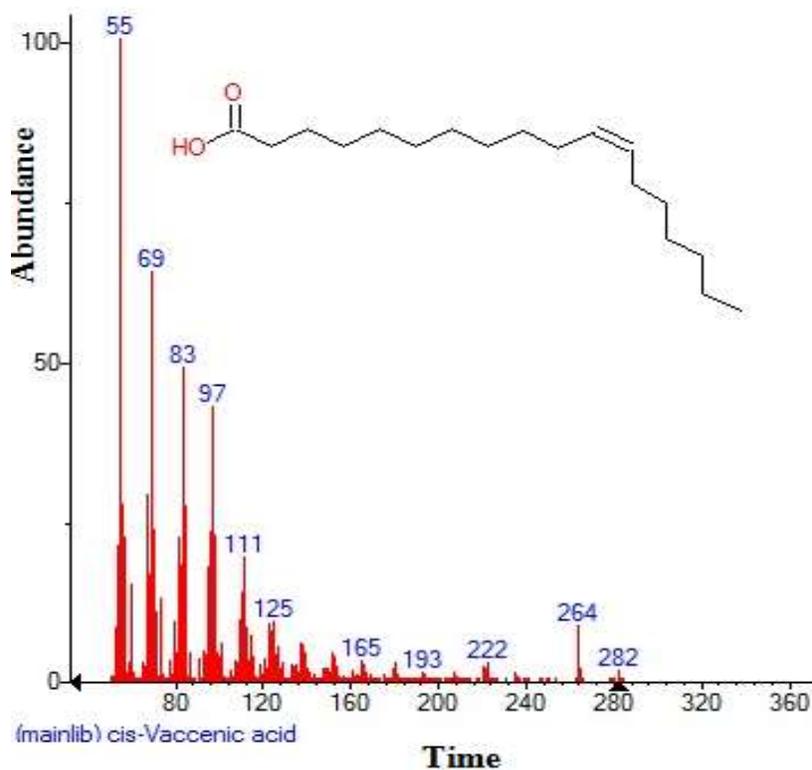


Figure 12. Structure of Cis-Vaccenic acid present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

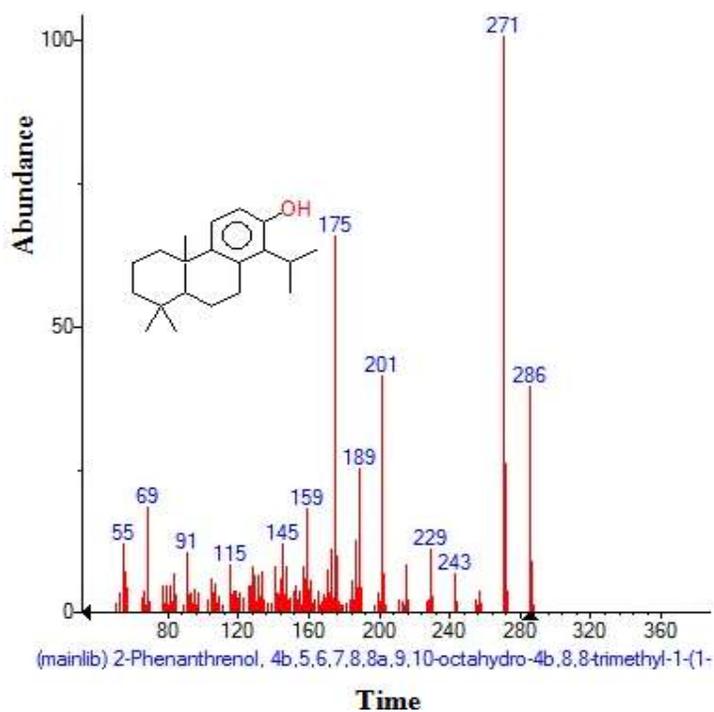


Figure 13. Structure of 2-Phenanthrenol,4b,5,6,7,8,8a,9,10-octahydro-4b,8,8 trimethyl-1 present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

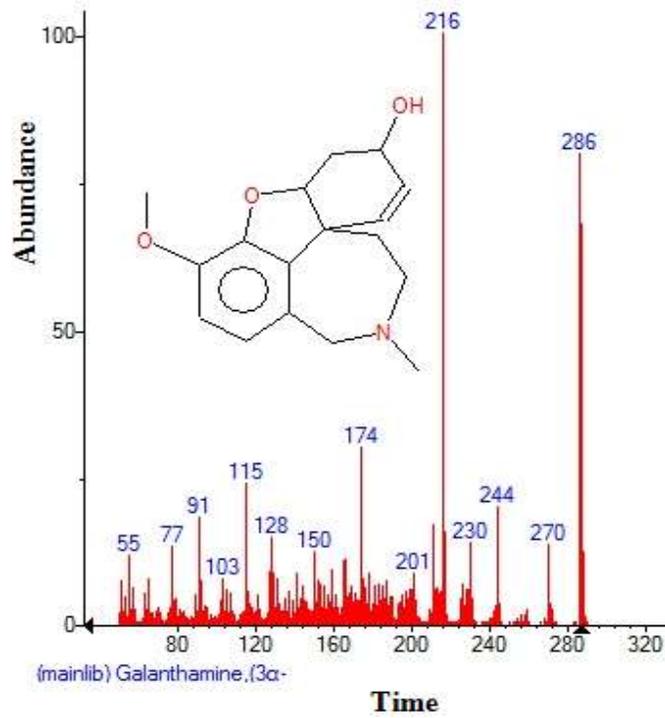


Figure 14. Structure of 2 Galanthamine present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

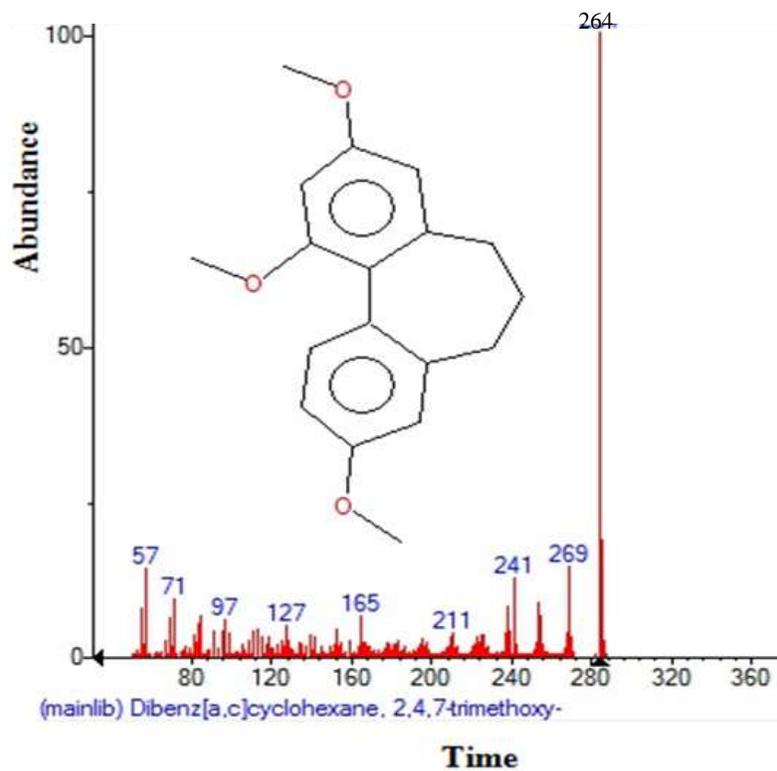


Figure 15. Structure of Dibenzo[a,c]cyclohexene, 2,4,7-trimethoxy present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

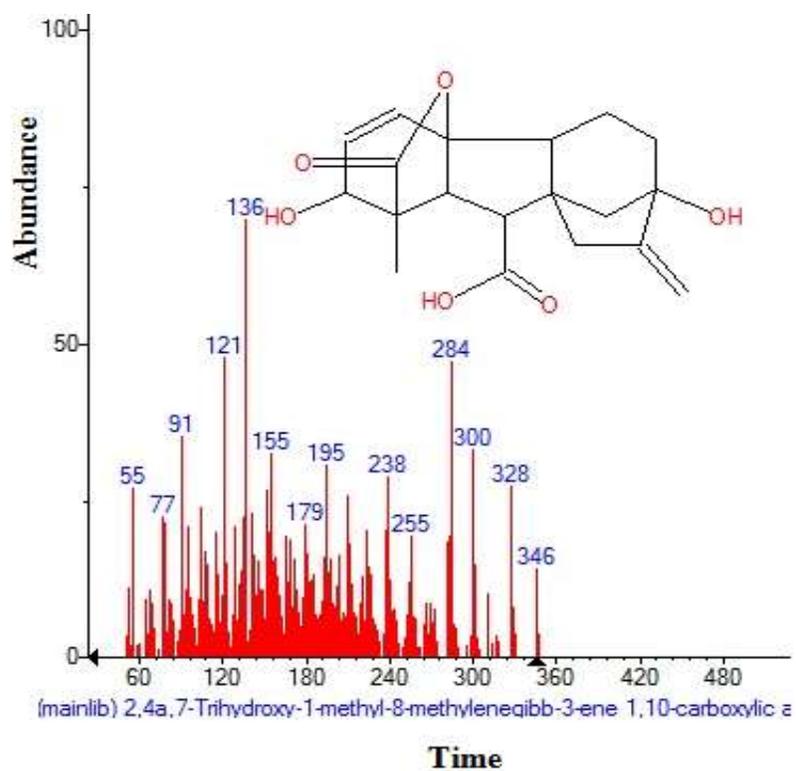


Figure 16. Structure of 2,4,7-Trihydroxy-1-methyl-8-methyleneqibb-3-ene 1,10-carboxylic acid. present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

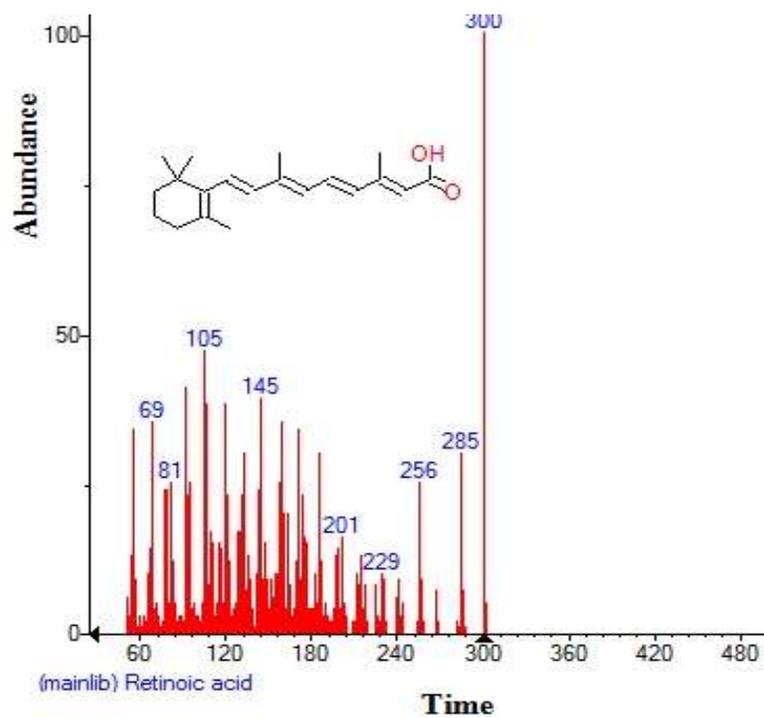
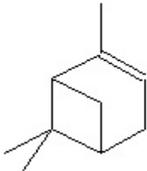
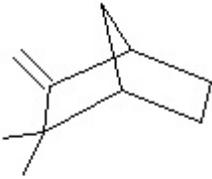


Figure 17. Structure of Retinoic acid present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

Table 1. Major phytochemical compounds identified in methanolic extract of *Rosmarinus officinalis* leaves.

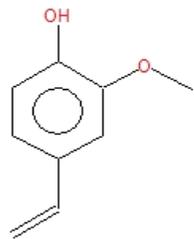
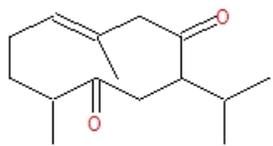
Serial No.	Phytochemical compound	RT(min)	Formula	Molecular weight	Exact mass	Chemical structure	MS Fragment ions	Pharmacological actions
1.	α -pinene	3.298	<u>C₁₀H₁₆</u>	136	136.1252		53,67,74,77,93,105,121,136	Antimicrobial against bacterial and fungal cells activities
2.	Camphene	3.476	<u>C₁₀H₁₆</u>	136	136.1252		53,67,79,93,107,121,136	Antimicrobial against <i>Escherichia coli</i> , <i>Staphylococcus aureus</i> and <i>Candida albicans</i> , but was not active against <i>Clostridium perfringens</i> up to the concentration of 100 g/ml. The significant antimicrobial and antioxidant activities of <i>R. minima</i> oil suggests that it could serve as a source for compounds with therapeutic potential.
3	Eucalyptol	4.317	<u>C₁₀H₁₈O</u>	154	154.13576		55,71,81,93,108,125,154	Eucalyptol, 1,8 cineole, is an essential oil present in large amounts in a variety of plants which is frequently used in the manufacture of cosmetics, to increase percutaneous penetration of drugs, as a nasal decongestant and anticough agent, in aromatherapy, and in dentistry (1-4). Eucalyptol has been used to treat bronchitis, sinusitis and chronic rhinitis and also for the treatment of asthma

methanolic extract of *Rosmarinus officinalis* leaves proved the presence of Alkenes, Aliphatic fluoro compounds, Alcohols, Ethers, Carboxylic acids, Esters, Nitro compounds, Alkanes, Aldehydes,

Ketones compounds which shows major peaks at 918.12, 1028.06, 1101.35, 1141.15, 1313.52, 1361.74, 1732.08 (Table 2; Figure 18). Among the identified phytocompounds have the property of

antioxidant and antimicrobial activities (He, 2010 and Deus-de-Oliveira, 2011). Plant based antimicrobials have enormous therapeutic potential as they can serve the purpose with

Table 1. Cont'd.

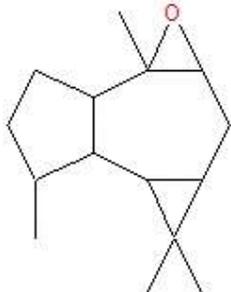
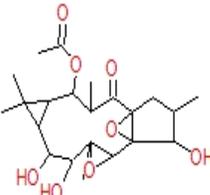
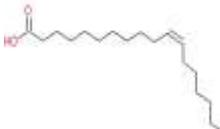
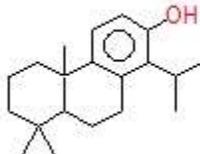
4.	2-Methoxy-4-vinylphenol	7.928	<u>C₉H₁₀O₂</u>	150	150.06808		51,77,89,107,121,135	Anti-Oxidant and Anti-inflammatory
5.	1-Oxaspiro[4,5]deca-3,6-diene,2,6,10,10-tetramethyl	9.862	<u>C₁₃H₂₀O</u>	192	192.151415		53,65,77,93,121,136,149,174,192	New chemical compound
6.	3-(N,N-Dimethyllaurylammonio)propanesulfate	10.348	<u>C₁₇H₃₇N₅O₃</u>	335	335.249414		69,84,97,108,128,140,152,165,179,194,213	New chemical compound
7.	Neocurdione	11.063	<u>C₁₅H₂₄O₂</u>	236	236.17763		55,69,82,109,137,167,180,193,236	Anti-viral, Anti-bacteria and Anti-Tumor activity

lesser side effects. Continued further exploration of plant derived antimicrobials is needed today.

Boutekdjiret et al., (2003) studied the constituents of rosemary essential oil from Algeria

They reported 1, 8-cineole, camphor, β -pinene, and α -Pinene as the major constituents in

Table 1. Cont'd.

8.	Isoaromadendrene epoxide	12.179	<u>C₁₅H₂₄O</u>	220	220.182715		55,67,93,107,135,149,162,187,220	Antibacterial activity and antioxidant activity
9.	1b,4a-Epoxy-2H-cyclopenta[3,4]cyclopropa[8,9]cycloundec.	12.356	<u>C₂₂H₃₂O₈</u>	424	424.209419		69,97,109,123,137,152,165,181,237,261,295,317,346	New chemical compound
10.	Cis-Vaccenic acid	16.940	<u>C₁₈H₃₄O₂</u>	282	282.25588		55,69,83,97,111,125,165,193,222,264,282	anti-inflammatory
11.	2-Phenanthrenol, 4b,5,6,7,8,8a,9,10-octahydro-4b,8,8-trimethyl-1	18.696	<u>C₂₀H₃₀O</u>	286	286.226999		55,69,91,115,145,159,175,189,201,229,243,271,286	Antimicrobial activity

the oil. Viuda-Martos et al., (2007) investigated chemical composition of the essential oil of anther sample of rosemary leaves from Spain. The major

constituents identified were α -pinene, camphor, 1,8-cineole and camphene. Chemical composition of essential oils of rosemary from various

geographic origins in Iran were determined by GC-MS. The main components detected in the oils were: α -pinene, 1, 8-cineole, camphene, camphor,

Table 1. Cont'd.

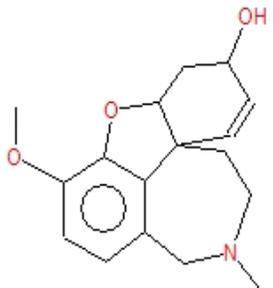
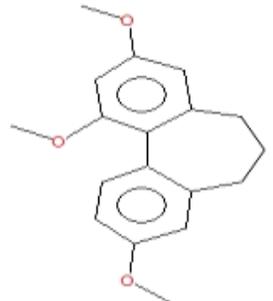
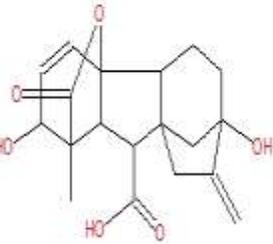
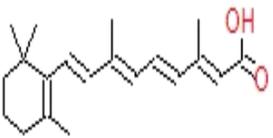
12.	Galanthamine	20.047	<u>C₁₇H₂₁NO₃</u>	287	287.152143		55,77,91,103,115,128,150,174,201,216,230,244,270,286	Galantamine hydrobromide is a tertiary alkaloid drug that has been developed and approved in a number of countries including the USA and several countries in Europe as a treatment for mild-to-moderate Alzheimer's disease (AD).
13.	Dibenz[a,c]cyclohexane,2,4,7-trimethoxy	20.682	<u>C₁₈H₂₀O₃</u>	284	284.141245		57,71,97,127,165,211,241,269,284	New chemical compound
14.	2,4a,7-Trihydroxy-1-methyl-8-methylenejibb-3-ene. 1,10-carboxylic acid.	20.911	<u>C₁₉H₂₂O₆</u>	346	346.141623		55,77,91,121,136,155,179,195,238,255,284,300,328,346	New chemical compound
15.	Retinoic acid	22.387	<u>C₂₀H₂₂O₂</u>	300	300.208931		69,81,105,145,201,229,256,285,300	Antibacterial activity and antioxidant activity

Table 1. Cont'd.

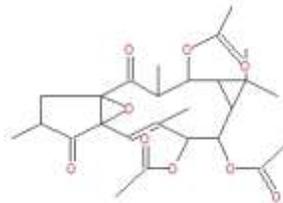
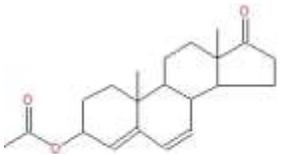
16.	7,8,12-Tri-O-acetyl-3-desoxy-ingol-3-one	26.512	<u>C₂₆H₃₄O₉</u>	490	490.220284		60,91,137,163,181,208,243,311,328,371,431,448,490	Antibacterial activity and antioxidant activity
17.	4,6-Androstadien-3β-ol-17-one,acetate	27.680	<u>C₂₁H₂₈O₃</u>	328	328.203844		55,79,119,149,197,232,268,286,328	antioxidant activity and antibacterial activity

Table 2. FT-IR peak values of *Rosmarinus officinalis* methanolic leaves extract.

No.	Peak (Wave number cm ⁻¹)	Intensity	Bond	Functional group assignment	Group frequency
1.	918.12	79.224	C-H	Alkenes	675-995
2.	1028.06	65.553	C-F stretch	Aliphatic fluoro compounds	1000-10150
3.	1101.35	71.963	C-O	Alcohols, Ethers, Carboxylic acids, Esters	1050-1300
4.	1141.15	75.676	C-O	Alcohols, Ethers, Carboxylic acids, Esters	1050-1300
5.	1240.23	76.745	C-O	Alcohols, Ethers, Carboxylic acids, Esters	1050-1300
6.	1313.52	80.282	NO ₂	Nitro Compounds	1300-1370
7.	1361.74	79.555	C-H	Alkanes	1340-1470
8.	1606.70	82.791	-	Unknown	-
9.	1732.08	81.440	C=O	Aldehydes, Ketones, Carboxylic acids, Esters , Alkenes	1690-1760
10.	2848.86	83.071	-	Unknown	-
11.	2918.30	78.923	C-H	Alkanes	2850-2970

myrcene and broneol (Jamshidi et al., 2009). GC and GC-MS analysis of oils from rosemary leave samples from India revealed the presence of

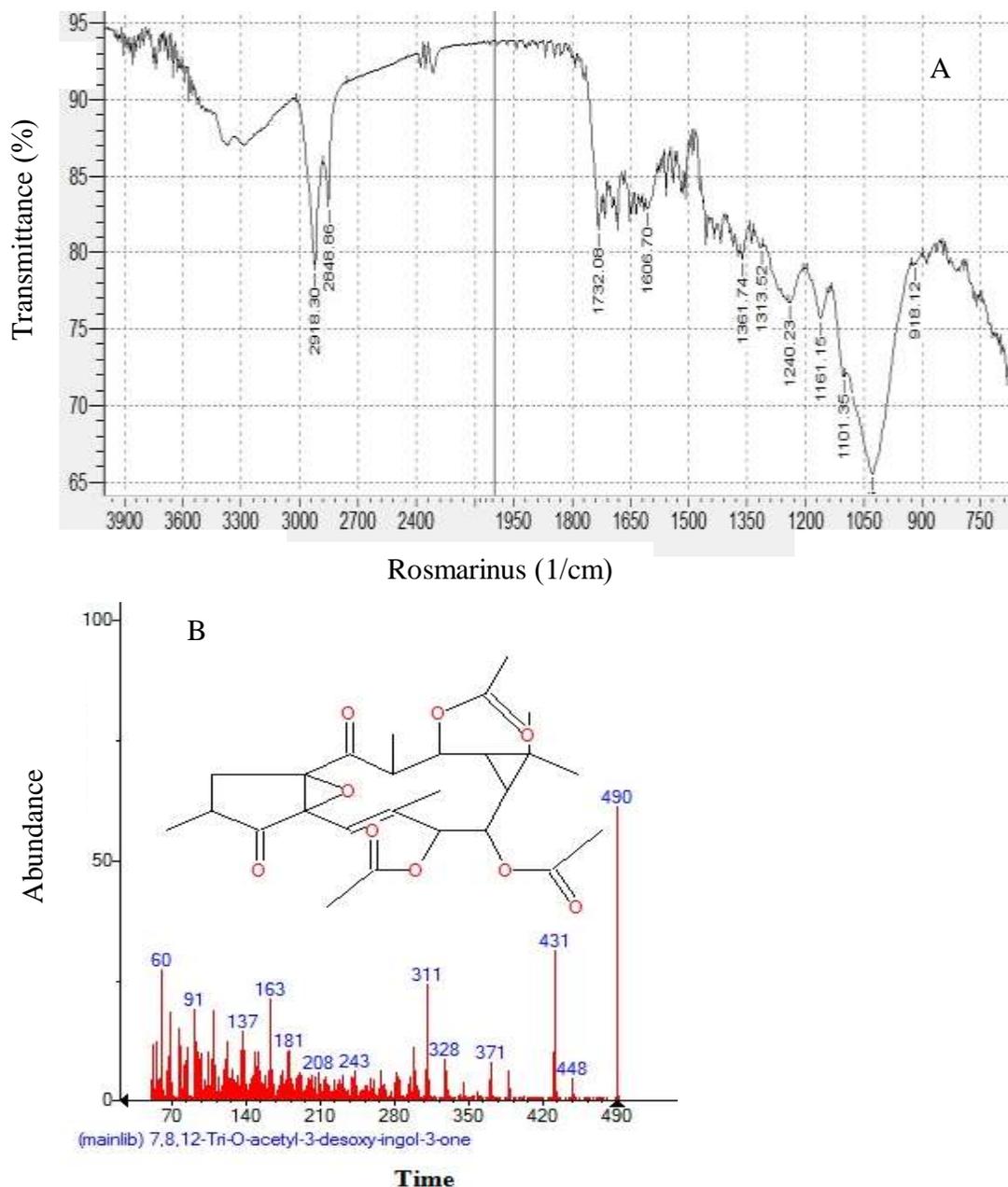


Figure 18. (A) FT-IR profile of leaf extract of *Rosmarinus officinalis*. **(B)** Structure of 7,8,12-Tri-O-acetyl-3-desoxy-ingol-3-one present in the leaf extract of *Rosmarinus officinalis* using GC-MS analysis.

camphor, 1,8-cineole and α -pinene as major constituents in the oils (Ram et al., 2011)

Conclusion

Rosmarinus officinalis is native plant of Iraq. It contains chemical constitutions which may be useful for various herbal formulation as anti-inflammatory, analgesic, antipyretic, cardiac tonic and antiasthmatic.

Conflict of Interest

The authors have not declared any conflict of interest.

ACKNOWLEDGEMENT

Authors thank Dr. Abdul-Kareem Al-Bermani, Lecturer, Department of Biology, for valuable suggestions and encouragement.

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